# Asymmetric Cyclopropanation Using New Chiral Auxiliaries Derived from D-Fructose 

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#### Abstract

Acetals of $\alpha, \beta$-unsaturated aldehydes with 3-O-alkylated derivatives of $1,2-O$-isopropylidene- $\beta$-Dfructopyranose and $1,2-O$-isopropylidene- $\beta$-D-psicopyranose, which are readily available from D-fructose, were cyclopropanated with $\mathrm{Et}_{2} \mathrm{Zn}$ and $\mathrm{CH}_{2} \mathrm{I}_{2}$ with good diastereoselectivity. The effects of structure of the acetals on enantioselectivity were examined.


## Introduction

The enantioselective methylenative cyclopropanation methods for prochiral alkenes are mainly classified into two categories: addition of a methylene group to a chiral alkene (i.e. alkenes with chiral auxiliaries) ${ }^{1}$ and to an achiral alkene in the presence of chiral catalysts. ${ }^{2}$ Although the latter method is highly appealing, its synthetic potential has not been fully developed to a synthetically useful level.

The chiral auxiliaries include oxazolidinones, ${ }^{1 \mathrm{a}}$ oxazolidines, ${ }^{1 \mathrm{~b}}$ carbohydrates, ${ }^{1 \mathrm{c}}$ phenylmenthyl alcohols, ${ }^{1 \mathrm{~d}}$ 1,4-di-O-benzyl-L-threitol, ${ }^{1 \mathrm{e}}\left(\mathrm{S}, \mathrm{S}\right.$ )-hydrobenzoin, ${ }^{1 \mathrm{f}}$ and tartaric esters, ${ }^{18}$ etc. These reactions are carried out with the reagents prepared from $\mathrm{Zn}-\mathrm{Cu}, \mathrm{Zn}-\mathrm{Ag}$ couple ${ }^{1 \mathrm{~h}}$ or $\mathrm{Et}_{2}-$ $\mathrm{Zn}^{1 i}$ with $\mathrm{CH}_{2} \mathrm{X}_{2}(\mathrm{X}=\mathrm{Br}$ or I). Among the synthetic methods described, the most widely recognized methodology for the stereoselective generation of cyclopropanes are the Simmons-Smith reaction and its various modifications. ${ }^{3}$

Recent studies of asymmetric syntheses using carbohydrate derivative encouraged us to explore the possibility of the asymmetric Simmons-Smith reaction. ${ }^{4}$ The design of a new chiral auxiliary for this reaction was

[^0]based on the observation that oxygen atoms in a suitably substituted carbohydrate derivative proximal to the alkene could undergo prior coordination of the zinc atom by the reagent, which can direct the attack by the reagent. ${ }^{5}$ In this paper, asymmetric cyclopropanation reactions of $\alpha, \beta$-unsaturated carbonyl compound linked to new chiral auxiliaries, 3-O-alkylated derivatives of 1,2 -$O$-isopropylidene- $\beta$-D-fructopyranose 1 and 1,2- $O$-isopro-pylidene-3- $O$-( $p$-phenylbenzyl)- $\beta$-D-psicopyranose 2 , which are readily available from D-fructose and reusable (vide infra), were attempted with the hope of stereoselective delivery of the reagent to only one side of the double bond.


1


2

## Results and Discussion

Synthesis of Chiral Auxiliaries. The starting material 1,2:4,5-di- $O$-isopropylidene- $\beta$-D-fructopyranose (4) was readily prepared from D-fructose (3) in acetone. ${ }^{6}$ Alkylation of 4 with an excess of various alkyl halides, followed by treatment of the resulting mixture with $80 \%$ aqueous acetic acid at room temperature, gave $3-O$-alkylprotected diol 6 in good yield as shown in Scheme 1. The results with various alkyl halides are summarized in Table 1.

Also, its C-3 epimer 17 was prepared as described in Scheme 2. Oxidation of the alcohol 4 with pyridinium chlorochromate ${ }^{7}$ gave the corresponding ketone in $65 \%$ yield. Subsequent reduction with $\mathrm{LiAlH}_{4}$ at $-78{ }^{\circ} \mathrm{C}$ provided the epimeric alcohol $16\left\{[\alpha]_{D}-110.9^{\circ}\right.$ (c 0.24 , acetone) $\}$ in $93 \%$ yield, O-alkylation of which with

[^1]Scheme $1^{a}$


5
$\xrightarrow{b}$
$$
\xrightarrow{c}
$$


6
${ }^{a} \mathrm{Key:}$ (a) $0.5 \% \mathrm{H}_{2} \mathrm{SO}_{4}$ acetone, $2 \mathrm{~h}, \mathrm{rt}, 45 \%$; (b) $\mathrm{NaH}, \mathrm{R}_{1} \mathrm{X}, \mathrm{DMF}$, $1 \mathrm{~h}, \mathrm{rt}$; (c) $80 \%$ aqueous $\mathrm{AcOH}, 12 \mathrm{~h}$, rt (or $3-4 \mathrm{~h}, 60^{\circ} \mathrm{C}$ ).
p-phenylbenzyl chloride followed by deacetalization with $80 \%$ aqueous acetic acid produced the diol $17\left\{[\alpha]_{\mathrm{D}}-80.9^{\circ}\right.$ (c 0.216 , acetone) $\}$ in $79 \%$ yield for the two steps.

Synthesis of Aldehydes. Various trans aldehydes were synthesized as shown in Scheme 3. And the diethyl acetals were prepared from the corresponding enals by reaction with triethyl orthoformate in the presence of a catalytic amount of ammonium nitrate in ethanol. ${ }^{8}$
Synthesis of Chiral Acetal Derivatives. The acetal derivatives of $\beta$-D-fructopyranose (27A and 27B) were obtained by transacetalization of the diethyl acetals of trans enals with the diol 6 in the presence of pyridinium tosylate as shown in Scheme 4.
Interestingly, the endo isomer 27A was a major product in all cases, with the ratios of 27A to $27 B$ being $1.1: 1$ to $1.8: 1$ as shown in Table 2. Fortunately, the mixture of 27A and 27B were easily separated by flash chromatography.
The relative configurations in the structure of the acetal 27 (i.e. A or B) were established by NOE experiments. ${ }^{9}$ For example, in the case of the cinnamylidene-$O$-benzyl derivative 28 , the $\mathrm{H}_{7}$ proton in the endo isomer (A) showed $2 \%$ NOE enhancement upon irradiation of the proton $\mathrm{H}_{4}$, while no such effect was seen between the two protons in the exo isomer (B). Furthermore, upon irradiation of the proton $\mathrm{H}_{7}$ in the endo (A) and exo (B) isomers of the acetal $\mathbf{2 8}$, respectively, NOE enhancement for $\mathrm{H}_{3}$ was detected only in the exo isomer (B) (Figure 1). These experiments unequivocally revealed that $\mathrm{H}_{7}$ and $\mathrm{H}_{4}$ in the endo isomer ( $\mathbf{A}$ ) and $\mathrm{H}_{7}$ and $\mathrm{H}_{3}$ in the exo isomer ( $\mathbf{B}$ ) were in close proximity, as depicted in Figure 1. The same analysis could be applied to the acetal 37 (Figure 1). Additionally, force field calculations have shown the most plausible conformation, which is consistent with these NMR data, as the one in which the endo acetal isomers adopt a chair conformation with the $\beta$-stryryl group stretching away from the concave space (Figure 1). More importantly, the $\beta$-styryl group in the endo acetal is approximately $\pi$-stacked with the $O$-benzyl group ( $R_{1}$ in the structure of $\mathbf{2 7 A}$ ) in the conformer at the global minimum, while the two groups in the global minimum conformer of the exo acetal are quite independent of each other, which has some implication in why the endo acetals gave higher enantioselectivity than the exo acetals as we shall see later.

The acetal derivatives of $\beta$-D-psicopyranose (41A and 41B) were prepared by the method shown in Scheme 5.

[^2]After this reaction, the approximately $1: 1$ mixture of the endo and exo isomers was readily separated by flash chromatography $\left\{41 \mathrm{~A},[\alpha]_{\mathrm{D}}-160.6^{\circ}\right.$ (c 0.127 , acetone); $41 B,[\alpha]_{D}-60.0^{\circ}$ (c 0.22 , acetone) $\}$. The stereochemistry of the acetal moiety in 41A and 41B was again confirmed by NOE measurements (Figure 2). Especially, upon irradiation of $\mathrm{H}_{4}$ in each case, only the endo isomer 41A showed NOE enhancement at $\mathrm{H}_{7}$ by $7.1 \%$. Supportive to this conclusion was the molecular modeling study, in which the global minimum structures of the pyranose moiety adopt boat conformations in which the severe 1,2interaction of the $p$-phenylbenzyloxy group with the adjacent 1,3 -dioxolane ring is more or less minimized by the ring deformation and placing the $p$-phenylbenzyl unit completely away from the sugar ring (Figure 2).

As shown in Scheme 6, the acetal derivatives of cis aldehydes with $\beta$-D-fructopyranose 43 were obtained by Lindlar reduction of the corresponding acetylenic aldehyde acetals with $\beta$-D-fructopyranose, which were prepared by treatment of the diol 6 with the diethyl acetals of the corresponding acetylenic aldehyde in the presence of pyridinium tosylate. The results are summarized in Table 3.

The endo nature of the acetal 45 was confirmed by 1.8\% NOE enhancement of $\mathrm{H}_{7}$ on irradiation of $\mathrm{H}_{4}$ (Figure 3).

Cyclopropanation. Various conditions for cyclopropanation of trans-cinnamylidene- $O$-benzyl- $\beta$-D-fructopyranose (28A) as a model substrate using $\mathrm{Et}_{2} \mathrm{Zn} / \mathrm{CH}_{2} \mathrm{I}_{2}$ were examined in the Simmons-Smith type reaction (Scheme 7). The corresponding chloromethyl zinc reagent, which is known to afford a higher yield than the iodomethyl zinc analog because of better stability, ${ }^{10}$ gave lower yields of cyclopropanation although it gave almost the same enantioselectivity. A samarium-based cyclopropanation ${ }^{12}$ as an alternative was not successful (Table 4). For cyclopropanation, a $2: 1$ stoichiometry of diiodomethane to $\mathrm{Et}_{2} \mathrm{Zn}$ was used throughout this study unless stated otherwise. This was followed by hydrolysis of the chiral auxiliary, which could be recovered in high yield without loss of chirality, and subsequent reduction of the aldehyde functionality to give cyclopropanemethanol, of which ee and absolute configurations were determined. The results are summarized in Table 4.

For cyclopropanation, 1,2-dichloroethane and toluene were found to be superior solvents for this reaction. The cyclopropanation in toluene or 1,2 -dichloroethane gave similar results in terms of enantioselectivity. A cosolvent system (e.g., toluene-hexane or 1,2-dichloroethanehexane) gave almost the same enantioselectivity but with lower yields. The failure of hexane to act as an efficient medium for cyclopropanation was apparently due to the insolubility of the starting material. In contrast, ethereal solvent (e.g., THF) was undesirable, since the rate of reaction dramatically decreased.

[^3]Table 1. Preparation of $\mathbf{1 , 2 - O}$-Isopropylidene-3-O-alkyl- $\beta$-D-fructopyranose 6 from 4

| entry | $\mathrm{R}_{1} \mathrm{X}$ | product (6, |
| :---: | :---: | :---: |
| 1 | BnBr | Bn |
| 2 | $\mathrm{CH}_{3} \mathrm{I}$ | $\mathrm{CH}_{3}$ |
| 3 | o- $\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{Br}$ | $o-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ |
| 4 | $m-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{Br}$ | $m-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ |
| 5 | $p-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{Br}$ | $p-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ |
| 6 | $p-\mathrm{MeOC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{Cl}$ | $p-\mathrm{MeOC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ |
| 7 | $p-t-\mathrm{BuC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{Br}$ | $p-t-\mathrm{BuC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ |
| 8 | $p-\mathrm{PhC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{Cl}$ | $p \cdot \mathrm{PhC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ |
| 9 | $2-\mathrm{NpCH}_{2} \mathrm{Br}$ | $2-\mathrm{NpCH}_{2}$ |
|  | Scheme $\mathbf{2}^{\boldsymbol{a}}$ |  |
| $\underline{a, b}$ |  |  $R=p-\mathrm{PhC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ <br> 17 |

${ }^{a}$ Key: (a) $\mathrm{PCC}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 3 \mathrm{~h}, 65 \%$; (b) $\mathrm{LiAlH}_{4}, \mathrm{THF},-78{ }^{\circ} \mathrm{C}, 1$ $\mathrm{h}, 93 \%$; (c) $\mathrm{NaH}, \mathrm{DMF}, p-\mathrm{PhC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2} \mathrm{Cl}, \mathrm{rt}, 1 \mathrm{~h}$; (d) $80 \%$ aqueous $\mathrm{AcOH}, \mathrm{rt}, 12 \mathrm{~h}, 79 \%$.

## Scheme $3^{a}$


${ }^{a}$ Key: (a) $\mathrm{PCC}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{rt}, 80-85 \%$; (b) $\mathrm{Ph}_{3} \mathrm{P}=\mathrm{CHCO}_{2} \mathrm{Et}$, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, 1 \mathrm{~h}, 89 \%$; (c) DIBAL, benzene, rt, $12 \mathrm{~h}, 81 \%$; (d) NaH, $\mathrm{BnBr}, \mathrm{DMF}, \mathrm{rt}, 1 \mathrm{~h}, 38 \%$; (e) $\mathrm{LiC} \equiv \mathrm{CCH}_{2} \mathrm{OTBDPS}, \mathrm{HMPA},-78$ ${ }^{\circ} \mathrm{C}, \mathrm{rt}, 5 \mathrm{~h}, 80 \%$; (f) $n$ - $\mathrm{Bu} 4 \mathrm{NF}, \mathrm{THF}, \mathrm{rt}, 1 \mathrm{~h}, 89 \%$; (g) $\mathrm{HC}(\mathrm{OEt})_{3}$, $\mathrm{NH}_{4} \mathrm{NO}_{3}, \mathrm{EtOH}, \mathrm{rt}, 12 \mathrm{~h}$.

## Scheme 4

6


PPTS, benzene, $\Delta, 6-8 \mathrm{~h}$


Lowering the reaction temperature dramatically enhanced the enantioselectivity in these reactions. Especially, the enantioselectivity improved to $90 \%$ in toluene at $-15{ }^{\circ} \mathrm{C}$. The enantiomeric excess of this reaction in 1,2-dichloroethane at $-20^{\circ} \mathrm{C}$ was $12 \%$ higher than at 0 ${ }^{\circ} \mathrm{C}$ in the same solvent, but at a sacrifice to the yield. Also, the reaction time affected the enantioselectivity:

The shorter reaction time yielded higher enantioselectivity but resulted in lower yields as shown in Table 4. There may be some incompatibility of the starting materials with the reagent system. The direct evidence for this will be elaborated later.

Secondly, the face-blocking effect of $R_{1}$ in the chiral sugar moiety was examined (Scheme 8, Table 5): The alkyl group on the sugar moiety $\left(\mathrm{R}_{1}\right)$ dramatically affected enantioselectivity in these reactions: The size of $R_{1}$ was proportional to enantioselectivity ( $\mathrm{R}_{1}=\mathrm{Me} \ll \mathrm{Bn}<$ $p-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}<p-\mathrm{PhC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ ). However, the alkyl substituents ( X ) on the phenyl ring of $\mathrm{R}_{1}$ had virtually no effect on enantioselectivity with the order of $p-\mathrm{PhC}_{6} \mathrm{H}_{4}$ $\mathrm{CH}_{2}>\mathrm{Bn} \cong m-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}>o-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2} \cong p-t$ $\mathrm{BuC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}>2-\mathrm{NpCH}_{2}$. This could be attributed to the steric repulsion between the bulky alkyl groups on the phenyl ring of the sugar moiety and the 5 -membered acetal moiety. Consequently, the phenyl ring with the substituent $X$ lies far from the oxygen atoms of the acetal moiety, minimizing the blocking chance of the benzyl group against either diastereotopic face of the double bond.

However, there is still the possibility that prior coordination of the reagent to the ethereal oxygen atom, especially the acetal oxygen atoms, precedes the cyclopropanation. Thus, as the size of the substituent $X$ on the phenyl ring becomes larger, the possibility in which the Simmons-Smith reagent indiscriminately coordinates all of the acetal oxygen atoms increases, which would cause lower enantioselectivity. This is supported by the case of 3-O-( $p$-methoxybenzyl)-substituted auxiliary 33A, with which the enantioselectivity was worse than expected, presumably due to the oxophilicity of the zinc reagent toward the methoxy oxygen. In other words, since the methoxy oxygen of the phenyl group would coordinate with the Simmons-Smith reagent, it would allow for transfer of the incipient carbenoid to either side of the double bond.

The fact that the enantioselectivity was higher in the case of $p$-phenylbenzyl-substituted derivatives 35A can be attributed to $\pi$-stacking between the double bond and the phenyl group, as mentioned before. In other words, since one face was more efficiently blocked than other one, the facial selectivity increased (Scheme 8, Table 5).
As shown Scheme 9, with the p-phenylbenzyl group in the chiral auxiliary fixed, attention was focused on the general applicability of the reaction by varying the enals, i.e. the substituent $\left(R_{2}\right)$ on the olefin, to determine the scope of the reaction as shown Table 6.

While almost the same high enantioselectivity was obtained when $R_{2}$ was phenyl or $\beta$-phenethyl, the enantioselectivity dramatically decreased in the case when $R_{2}$ was a benzyloxy group. This may be due to the oxophilicity of zinc atom toward the oxygen atom of the benzyloxy moiety via chelation, as mentioned before.

Table 2. Preparation of the Acetals of Trans Enal 27 from 3-O-Alkyl- $\beta$-d-fructopyranose 6

| products |  |  |  |  |  | $[\alpha]_{D}$ in acetone |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{R}_{1}$ | $\mathrm{R}_{2}$ | $\mathrm{R}_{3}$ | no. | endo:exo | yield (\%) ${ }^{\text {a }}$ | endo (A) | exo (B) |
| Bn | Ph | H | 28 | 1.8:1 | 90 | $-126.7^{\circ}$ ( $\mathrm{c}^{0.105}$ ) | $-65.9^{\circ}$ ( $c 0.135$ ) |
| $\mathrm{CH}_{3}$ | Ph | H | 29 | 1.6:1 | $49^{\text {b }}$ | $-85.1^{\circ}(\mathrm{c} 0.047)$ |  |
| $o-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | Ph | H | 30 | 1.5:1 | 89 | $-112.7^{\circ}(\mathrm{c} 0.118)$ | $-47.9^{\circ}$ ( c 0.192) |
| $m-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | Ph | H | 31 | 1.5:1 | 93 | $-123.8^{\circ}(c 0.084)$ | $-49.5^{\circ}$ ( $c^{0.2}$ ) |
| $p-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | Ph | H | 32 | 1.4:1 | 87 | $-132.0^{\circ}(c 0.05)$ | $-54.7^{\circ}(c 0.137)$ |
| $p-\mathrm{MeOC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | Ph | H | 33 | 1.6:1 | 88 | $-119.2^{\circ}(c 0.052)$ | -50.0 ${ }^{\circ}(\mathrm{c} 0.112$ ) |
| $p-t-\mathrm{BuC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | Ph | H | 34 | 1.2:1 | 90 | $-112.4^{\circ}\left(c^{\text {c }} 0.121\right)$ | $-54.4^{\circ}(c 0.103)$ |
| $p$ - $\mathrm{PhC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | Ph | H | 35 | 1.5:1 | 90 | $-128.6^{\circ}$ ( c 0.091) | $-56.9^{\circ}$ ( $c^{0.065}$ ) |
| $2-\mathrm{NpCH}_{2}$ | Ph | H | 36 | 1.4:1 | 88 | -141.2 ${ }^{\circ}$ ( 0.102 ) | $-60.0^{\circ}$ (c 0.1$)$ |
| $p-\mathrm{PhC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | $\mathrm{PhCH}_{2} \mathrm{CH}_{2}$ | H | 37 | 1.5:1 | 88 | $-54.2^{\circ}(c 0.059)$ | $-38.5{ }^{\circ}$ (c 0.039) |
| $p$ - $\mathrm{PhC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | $\mathrm{BnOCH}_{2}$ | H | 38 | 1.1:1 | 90 | $-94.5^{\circ}(c 0.055)$ | $-69.8^{\circ}(c 0.263)$ |
| $p$ - $\mathrm{PhC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | 39 | 1.7:1 | 80 | $-82.2^{\circ}$ (c 0.107) | $-99.7^{\circ}(\mathrm{c} 0.305)$ |
| $p-\mathrm{PhC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | $n-\mathrm{C}_{5} \mathrm{H}_{11}$ | H | 40 | 1.4:1 | 87 | $-87.4^{\circ}(\mathrm{c} 0.154)$ | $-76.9^{\circ}$ (c 0.09$)$ |

${ }^{a}$ The combined yield of endo and exo isomers. ${ }^{b}$ The yield of endo isomer only.



| $\mathrm{R}_{3}=\mathrm{PhCH}_{2} \mathrm{CH}_{2}$ : | $=\rho-\mathrm{PhC}$ | 4 (37) |
| :---: | :---: | :---: |
| $\begin{gathered} \text { A } \\ \text { NOE } \% \\ \hline \end{gathered}$ | protons | $\stackrel{\mathrm{B}}{\text { NOE. \% }}$ |
| 1.1 | $\mathrm{H}_{4} \rightarrow \mathrm{H}_{\text {a }}$ | 2.0 |
| 3.5 | $\mathrm{H}_{4} \rightarrow \mathrm{H}_{3}$ | - |
| 1.1 | $\mathrm{H}_{4} \rightarrow \mathrm{H}_{7}$ | no NOE |
| 1.6 | $\mathrm{H}_{7} \rightarrow \mathrm{H}_{9}$ | - |
| no NOE | $\mathrm{H}_{7} \rightarrow \mathrm{H}_{3}$ | - |

Figure 1.


Next, the reaction with the exo acetal isomers was examined (Scheme 10). In these cases, the blocking group ( $\mathrm{R}_{1}$ ) cannot effectively block either side of the double bond in contrast to the endo isomers. Consequently, the enantiomeric excess was lower than that in the case of the endo forms as expected, which is shown in Table 7.

Next, this Simmons-Smith reaction was applied to the cis-cinnamaldehyde endo acetal derivative (Scheme 11, Table 8). The cyclopropanation reaction of $43\left(\mathrm{R}_{2}=\mathrm{Ph}\right)$ with diethylzinc and diiodomethane at $0^{\circ} \mathrm{C}$ surprisingly
endo isomer (41A)
exo Isomer (41B)


NOE \% protons NOE \%

| - | $H_{4} \rightarrow H_{a}$ | 3.3 |
| ---: | :--- | :---: |
| 12.2 | $H_{4} \rightarrow H_{1}$ | - |
| 9.7 | $H_{4} \rightarrow H_{3}$ | 10.6 |
| 7.2 | $H_{4} \rightarrow H_{5}$ | 6.3 |
| 7.1 | $H_{4} \rightarrow H_{7}$ | no NOE |
| 3.4 | $H_{7} \rightarrow H_{5}$ | - |
| 14.5 | $H_{7} \rightarrow H_{9}$ | 30 |

Figure 2.
Scheme $\mathbf{6}^{\boldsymbol{a}}$

${ }^{a}$ Key: (a) $\mathrm{R}_{2} \mathrm{C} \equiv \mathrm{CCH}(\mathrm{OEt})_{2}$, PPTS, benzene, $\Delta$, 8 h ; (b) Lindlar cat., $\mathrm{H}_{2}, \mathrm{MeOH}$.
gave a 2:1 mixture of an endo acetal of cis-cyclopropanecarboxaldehyde (56) and also an exo acetal of transcyclopropanecarboxaldehyde ( $\mathbf{5 7}$ ) $\left(\mathrm{R}_{2}=\mathrm{Ph}\right)$ in $96 \%$ yield. The ratio of products was determined from the integration of the benzylic methine protons (on the 3-membered rings) in ${ }^{1} \mathrm{H}$ NMR after isolation [cis $\delta 2.35$ (m), trans $\delta$ $2.10(\mathrm{~m})$ ]. Treatment of the mixture with $80 \%$ aqueous acetic acid gave a $2: 1$ mixture of cis- and trans-2phenylcyclopropanecarboxaldehydes in $93 \%$ yield. Further reaction of the resulting mixture with sodium cyanide, manganese dioxide, acetic acid, and methanol at room temperature afforded ( $1 R, 2 S$ )-methyl cis-2phenylcyclopropanecarboxylate (55\% yield) $\left\{[\alpha]_{D}-29.7^{\circ}\right.$ (c $0.175, \mathrm{CHCl}_{3}$ ); lit. ${ }^{13}$ ent-form $[\alpha]_{\mathrm{D}}+32.8^{\circ}$ (c 1.99, $\mathrm{CHCl}_{3}$ ) ) and ( $1 S, 2 S$ )-methyl trans-2-phenylcyclopropanecarboxylate ( $28 \%$ yield) $\left\{[\alpha]_{D}+173^{\circ}\left(c \quad 0.185, \mathrm{CHCl}_{3}\right.\right.$ ); lit. ${ }^{13}[\alpha]_{D}+324.7^{\circ}$ (c 1.24, $\mathrm{CHCl}_{3}$ ) $\}$ after flash chromatography. The absolute configuration of the esters was determined by the comparison of its specific rotation with a literature value. ${ }^{13}$ The enantiomeric excess was determined after conversion of the esters with diisobutyl-

Table 3. Preparation of Acetals of Cis Enal 43 from 3-O-Alkyl- $\beta$-D-fructopyranose 6

| $\mathrm{R}_{1}$ in 6 |  |  |  | product, $\mathbf{4 3}$ |
| :--- | :--- | :---: | :---: | :---: |
|  | $\mathrm{R}_{2}$ in acetal | product, $\mathbf{4 2}$ (yield, \%) | (yield, \%) | optical rotation |
| $p-\mathrm{PhC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | Ph | $\mathbf{4 4}(84)$ | $\mathbf{4 5}(82)$ | $[\alpha]_{\mathrm{D}}-14.5^{\circ}(c 0.069$, acetone $)$ |
| $\mathrm{PhCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}$ | $\mathbf{4 6}(70)$ | $\mathbf{4 7}(79)$ | $[\alpha]_{\mathrm{D}}-53.2^{\circ}(c 0.062$, acetone $)$ |  |



45
Figure 3.

aluminum hydride to ( $1 R, 2 S$ )-cis- and ( $1 S, 2 S$ )-trans-1-(hydroxymethyl)-2-phenylcyclopropane, respectively.
Since no isomerization was observed during the hydrolysis of the chiral auxiliary under various acidic conditions, the isomerization should only occur during cyclopropanation, during which period the coordinated starting material with Simmons-Smith reagent could isomerize via generation of an allylic cation which is further stabilized via resonance through the phenyl group. This is supported by the fact that the ciscyclopropylfructopyranose $43\left(\mathrm{R}_{2}=\mathrm{PhCH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}\right)$, which has no phenyl moiety at the allylic position, did not produce an isomerized product.

On the other hand, the endo and exo isomers of psicopyranose derivatives 41 (Scheme 12) gave lower enantiomeric excesses than those of the fructopyranose derivatives as shown in Table 9, which can be attributed to a reduced blocking effect of the $p$-phenylbenzyl group in the case of psicopyranose derivatives in contrast to fructopyranose derivatives (vide supra).

Conclusion. Among various isomeric compounds, the asymmetric cyclopropanation reaction provided good enantioselectivity (consistent attack on the same face) with high chemical yields especially with endo acetals of $1,2-O$-isopropylidene-3- $O$-( $p$-phenylbenzyl)- $\beta$-D-fructopyranose. However, the observed diastereoselectivity was not perfect. One of the reasons for this may be the tendency of the endo acetals, which gave better diastereoselectivity, to isomerize to the less suitable exo acetals during the cyclopropanation. For example, the transendo acetal 35A was cyclopropanated under standard conditions. Analysis of the resulting crude reaction mixture by NMR revealed it to be a diastereomeric mixture, which could be carefully separated into two fractions, each containing endo and exo acetals of cyclopropanated ( $R, R$ )-aldehydes, 51 and 65, in an approximately 13:1 ratio (Scheme 13). The diastereomeric excess of the endo acetals was found to be $76 \%$ as judged from the NMR of the diastereomeric endo acetal mixture [ $\delta$ $3.90(\mathrm{~d}, J=8.42 \mathrm{~Hz}$ ) (minor) and $\delta 3.83(\mathrm{~d}, J=8.40 \mathrm{~Hz})$
(13) Krieger, P. E.; Langrebe, J. A. J. Org. Chem. 1978, 43, 4447.
(major)]. This was further confirmed by the HPLC analysis of the ( $1 R, 2 R$ )-trans-1-(hydroxymethyl)-2-phenylcyclopropane ( $\mathbf{4 9 \text { ) ( } 7 7 \% \text { ee) obtained after hydrolysis }}$ of chiral auxiliary and subsequent reduction of the diastereomeric endo acetal mixture. However, the diastereomeric exo acetal mixture could neither be separated by HPLC analysis nor gave any discernible peak difference in NMR. Thus, direct hydrolysis of chiral auxiliary and subsequent reduction gave again ( $1 R, 2 R$ )-trans-1-(hydroxymethyl)-2-phenylcyclopropane (49) only in $24 \%$ ee (HPLC) (Scheme 13). One important implication here is that although the amount of isomerization was small (ca. 7\%), the endo-to-exo isomerization seemed to occur mainly after the cyclopropanation since the pure exo acetals 28B, 37B, and 40B gave ( $1 S, 2 S$ )-trans-1-(hy-droxymethyl)-2-aryl(or alkyl)cyclopropane in 33-54\% ee's.

## Experimental Section

General. Nuclear magnetic resonance spectra were recorded on 200 and 300 MHz spectrometers. Chemical shifts are reported as $\delta$ values in parts per million ( ppm ) from tetramethylsilane ( $\delta=0$ ) and reported as follows: chemical shifts, multiplicity ( $s$, singlet; d, doublet; t, triplet; q, quartet; m , multiplet; $b$, broad peaks), coupling constant, and integration or identified proton. Carbon-13 chemical shifts are reported in parts per million ( ppm ) from the center line of the chloroform- $d$ triplet ( 77.0 ppm ). Mass spectra were taken with an electron beam energy of $70 \mathrm{eV}(\mathrm{EI})$ unless otherwise stated.

Chiral columns for gas chromatographic analyses were Chiraldex B-TA ( $30 \mathrm{~m} \times 0.25 \mathrm{~mm}$ ) and G-TA ( $30 \mathrm{~m} \times 0.25$ mm ). Chiral HPLC columns used were Chiralcel OJ and OD. Melting points were uncorrected. Thin layer chromatographic analyses were performed on precoated the plates (silica gel 60 GF254, 0.25 mm thickness). For flash column chromatography was used silica gel Kieselgel 60 . When necessary, solvents and reagents were dried prior to use as follows: tetrahydrofuran was distilled from sodium benzophenone ketyl; hexane, benzene, and toluene were distilled from calcium hydride. Unless otherwise stated, the reagents were purchased and used without further purification.

Molecular Modeling. Energy minimization of compounds was carried out on a SG Iris Crimson workstation using Discover software. Energy minimization with the CVFF force field was carried out until the change in energy was less than $0.001 \mathrm{kcal} / \mathrm{mol}$ between two successive iterations. Molecular dynamics simulations were performed by the method provided by Discover. To perform conformational searching to find the global minimum, dynamics annealing runs are employed. Thus, the structure was heated to 900 K and gradually cooled down to 300 K , and then the molecule was allowed to equilibrate for 5 ps . A dynamics simulation of 10 ps was performed using a time step of 1 fs . The conformations were sampled from the collection of all conformers during the course of simulation and were minimized.
Preparation of $1,2: 4,5-\mathrm{Di}-O$-isopropylidene- $\boldsymbol{\beta}$-d-fructopyranose (4). To a suspension of D -fructose $(3,30 \mathrm{~g}, 166.5$ mmol ) and acetone ( 600 mL ) was added concd $\mathrm{H}_{2} \mathrm{SO}_{4}$ ( 2.916 mL ). The reaction mixture was stirred for 2 h at room temperature. The reaction mixture was cooled to $0^{\circ} \mathrm{C}$, and the reaction was quenched with $\mathrm{NaOH}(9.3 \mathrm{~g})$ in $\mathrm{H}_{2} \mathrm{O}(84 \mathrm{~mL})$. After evaporation of the acetone, the residue was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic phase was separated and washed with water and brine and then dried over $\mathrm{MgSO}_{4}$. After filtration and removal of solvents under reduced pressure, the

Table 4. Optimization of Cyclopropanation of Acetal of trans-Cinnamaldehyde 28A

| cyclopropanation ${ }^{\text {a }}$ |  |  | hydrolysis ${ }^{\text {c }}$ |  | reduction ${ }^{\text {d }}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| solvent | temp, ${ }^{\circ} \mathrm{C}$ (time, h ) | yield ${ }^{\text {b }}$ (\%) | reagent | yield ${ }^{\text {b }}(\%)$ | yield ${ }^{\text {b }}$ (\%) | $\mathrm{ee}^{e}(\%)$ |
| hexane | -78 $\rightarrow 0$ (20) | trace |  |  |  |  |
| taluene | $-40 \rightarrow 0$ (3) | $33{ }^{\text {f }}$ | $p-\mathrm{TsOH}$, aqueous EtOH | 82 | 93 | 88 |
| toluene | $-40 \rightarrow 0$ (20) | 72 | 80\% aqueous AcOH |  | $91^{5}$ | 75 |
| toluene | $-78 \rightarrow 0(48)$ | 81 | 80\% aqueous AcOH |  | $80^{g}$ | 72 |
| toluene ${ }^{h}$ | $-78 \rightarrow-30(24) \rightarrow-15(24)$ | $50 f$ | 80\% aqueous AcOH |  | $86^{g}$ | 90 |
| $\left(\mathrm{ClCH}_{2}\right)_{2}$ | $-20 \rightarrow 0(3)$ | $61{ }^{\prime}$ | $p-\mathrm{TsOH}$, aqueous EtOH | 81 | 90 | 82 |
| $\left(\mathrm{ClCH}_{2}\right)_{2}$ | $-20 \rightarrow 0$ (24) | 85 | 80\% aqueous AcOH |  | $88^{g}$ | $71^{i}$ |
| $\left(\mathrm{ClCH}_{2}\right)_{2}{ }^{h}$ | $-30 \rightarrow-20$ (48) | 54 | 80\% aqueous AcOH |  | $86^{9}$ | 83 |
| THF | $-78 \rightarrow 0(24)$ | trace |  |  |  |  |
| Tol:hex ${ }^{j}$ | 0 (65) | $50 f$ | 80\% aqueous AcOH |  | 608 | 76 |
| $\left(\mathrm{ClCH}_{2}\right)_{2}$ :hex ${ }^{k}$ | 0 (42) | $72^{f}$ | 80\% aqueous AcOH |  | $58^{8}$ | 74 |
| $\left(\mathrm{ClCH}_{2}\right)_{2}{ }^{l}$ | 0 (48) | $23 f$ | 80\% aqueous AcOH |  | $62^{g}$ | 70 |
| THF ${ }^{m}$ | $-78 \rightarrow 0(24)$ | trace |  |  |  |  |

${ }^{a} \mathrm{Et}_{2} \mathrm{Zn}: \mathrm{CH}_{2} \mathrm{I}_{2}=5 \mathrm{mmol}: 10 \mathrm{mmol}$ unless noted otherwise. ${ }^{b}$ Isolated yield by column chromatography unless noted otherwise. ${ }^{c}$ The reaction time for hydrolysis of chiral auxiliary was 15 min with $80 \% \mathrm{AcOH} / \mathrm{H}_{2} \mathrm{O}$ at $60^{\circ} \mathrm{C}$. Under this condition, the chiral auxiliary could be recovered in high yield without loss of chirality. ${ }^{d}$ The reaction time of reduction was 1 h with $\mathrm{NaBH}_{4}$ at room temperature. ${ }^{e} \mathrm{The}$ enantiomeric excess was based on HPLC analysis. (Daicel OJ, $10 \% i-\mathrm{PrOH}$ in hexane). The absolute configuration ( $1 R, 2 R$ ) of trans-1-(hydroxymethyl)-2-phenylcyclopropane was determined by the reported specific rotation. Lit. ${ }^{11}[\alpha]^{24} \mathrm{p}-46.6^{\circ}$ (c 2.64 , EtOH as $51.3 \%$ ee). $f$ The yields were based on ${ }^{1} \mathrm{H}$ NMR. $g$ Yield for two steps (hydrolysis-reduction). ${ }^{h} \mathrm{Et}_{2} \mathrm{Zn}_{2}: \mathrm{CH}_{2} \mathrm{I}_{2}=10 \mathrm{mmol}: 20 \mathrm{mmol}$. ${ }^{i}[\alpha]_{\mathrm{D}}-64.7^{\circ}$ (c $0.19, \mathrm{EtOH}) .{ }^{j}$ Toluene:hexane $=1: 2 .^{k} 1,2$-Dichloroethane:hexane $=1: 2 .{ }^{i} \mathrm{Et}_{2} \mathrm{Zn}: \mathrm{ClCH}_{2} \mathrm{I}=5 \mathrm{mmol}: 10 \mathrm{mmol}$. ${ }^{m}$ Reaction with $\mathrm{Sm}^{2} \mathrm{CH}_{2} \mathrm{I}_{2}$.

residue was recrystallized with ether/hexanes to afford the alcohol $4(19.6 \mathrm{~g}, 45 \%)$ as a white solid: $[\alpha]^{28} \mathrm{D}-156.6^{\circ}(\mathrm{c} 1.00$, acetone); mp 117.5-118 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ) $\delta 1.37$ ( $\mathrm{s}, 3 \mathrm{H}$ ), $1.44(\mathrm{~s}, 3 \mathrm{H}), 1.51(\mathrm{~s}, 3 \mathrm{H}), 1.53(\mathrm{~s}, 3 \mathrm{H}), 2.29(\mathrm{~d}, \mathrm{~J}$ $=8.25 \mathrm{~Hz}, 1 \mathrm{H}), 3.66(\mathrm{dd}, J=6.74 \mathrm{~Hz}, J=8.18 \mathrm{~Hz}$, $1 \mathrm{H}), 3.93-4.20(\mathrm{~m}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 25.90,26.23,26.37$, 27.90, 60.69, 70.34, 72.26, 73.29, 77.25, 104.49, 109.36, 111.80; IR $3460 \mathrm{~cm}^{-1}$; MS (EI) $m / e$ (relative intensity) $260\left(\mathrm{M}^{+}, 2\right), 245$ (34), 144 (25), 127 (20), 117 (74), 100 (33), 85 (50), 59 (87), 43 (100). Anal. Calcd for $\mathrm{C}_{12} \mathrm{H}_{20} \mathrm{O}_{6}$ : C, $55.37 ; \mathrm{H}, 7.74$. Found: C, 55.70; H, 7.84.

Preparation of 3-O-Benzyl-1,2-O-isopropylidene- $\boldsymbol{\beta}$-dfructopyranose (7). To a solution of 1,2:4,5-di- $O$-isopropy-lidene- $\beta$-D-fructopyranose ( $4,4.0 \mathrm{~g}, 15.4 \mathrm{mmol}$ ) in DMF ( 23 mL ) was added $60 \%$ sodium hydride ( $0.7 \mathrm{~g}, 17.5 \mathrm{mmol}$ ) at room temperature. After the solution was stirred for 0.5 h , benzyl bromide ( $1.92 \mathrm{~mL}, 16.2 \mathrm{mmol}$ ) was added to the above solution. The reaction mixture was stirred for 1 h at room temperature, diluted with water, and then extracted with ethyl acetate. The combined extracts were washed with water and brine, dried over $\mathrm{MgSO}_{4}$, and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel with 1:20 ethyl acetate-toluene to give 3-O-benzyl-1,2;4,5-di- $O$ -isopropylidene- $\beta$-D-fructopyranose ( $5,5.33 \mathrm{~g}, 99 \%$ ) as a white oil: $[\alpha]^{22.5}{ }_{\mathrm{D}}-89.6^{\circ}\left(c 2.22, \mathrm{CHCl}_{3}\right)$; syrup; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta$ $1.38(\mathrm{~s}, 3 \mathrm{H}), 1.41(\mathrm{~s}, 3 \mathrm{H}), 1.50(\mathrm{~s}, 3 \mathrm{H}), 1.53(\mathrm{~s}, 3 \mathrm{H}), 3.50(\mathrm{~d}$, $J=7 \mathrm{~Hz}, 1 \mathrm{H}), 3.87(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.99(\mathrm{dd}, J=0.8 \mathrm{~Hz}$, $J=13 \mathrm{~Hz}, 1 \mathrm{H}), 4.08(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.14(\mathrm{dd}, J=2.7 \mathrm{~Hz}$, $J=13 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.22 (ddd, $J=5.5 \mathrm{~Hz}, J=2.7 \mathrm{~Hz}, J=0.8 \mathrm{~Hz}$, $1 \mathrm{H}), 4.38$ (dd, $J=5.5 \mathrm{~Hz}, J=7.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.66(\mathrm{~d}, J=12 \mathrm{~Hz}$, $1 \mathrm{H}), 4.97(\mathrm{~d}, J=12 \mathrm{~Hz}, 1 \mathrm{H}), 7.29-7.40(\mathrm{~m}, 5 \mathrm{H})$. Anal. Calcd for $\mathrm{C}_{19} \mathrm{H}_{26} \mathrm{O}_{6}$ : C, 65.13; H, 7.48. Found: C, 65.37; H, 7.47.

A solution of the benzyl ether $5(1.04 \mathrm{~g}, 2.97 \mathrm{mmol})$ in $80 \%$ acetic acid in $\mathrm{H}_{2} \mathrm{O}(10 \mathrm{~mL})$ was stirred overnight at room temperature. After evaporation of solvent, the residue was purified by flash chromatography on silica gel with $1: 2$ acetone-hexane to give the benzyl ether diol 7 ( $827 \mathrm{mg}, 90 \%$ ) as a white solid. As an alternative, the crude benzyl ether could be directly hydrolyzed with $80 \%$ acetic acid in $\mathrm{H}_{2} \mathrm{O}$ to give the same product without lowering the yield: $[\alpha]_{D}-115.1^{\circ}$ (c 1.52, chloroform); mp 99-101 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}+\mathrm{D}_{2} \mathrm{O}\right) \delta$
1.42 (s, 3 H ), 1.48 ( $\mathrm{s}, 3 \mathrm{H}$ ), 3.67 (d, $J=9.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.74 (dd, $J=1.5 \mathrm{~Hz}, J=11.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.90-4.08(\mathrm{~m}, 5 \mathrm{H}), 4.71(\mathrm{~d}, J$ $=11.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.82(\mathrm{~d}, J=11.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.3-7.40(\mathrm{~m}, 5 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 26.15,26.81,63.55,69.69,71.30,71.82$, $75.33,76.52,105.56,111.95,127.94,128.09,128.65$; IR 3422 $\mathrm{cm}^{-1}$; MS (CI) m/e (relative intensity) $311(\mathrm{M}+1), 163$ (10), 117 (18), 103 (40), 91 (100), 59 (33). Anal. Calcd for $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{O}_{6}: \mathrm{C}, 61.92 ; \mathrm{H}, 7.14$. Found: C, 62.12; H, 7.15.

Using the above reaction protocol, the compounds listed below were prepared.
1,2-O-Isopropylidene-3-O-methyl- $\beta$-D-fructopyranose (8): $87 \%$ as an oily solid; $[\alpha]_{\mathrm{D}}-96.2^{\circ}$ (c $1.52, \mathrm{CDCl}_{3}$ ); ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.39(\mathrm{~s}, 3 \mathrm{H}), 1.46(\mathrm{~s}, 3 \mathrm{H}), 2.5(\mathrm{~b}, 2 \mathrm{H}), 3.39(\mathrm{dm}, J$ $=9.39 \mathrm{~Hz}, 1 \mathrm{H}), 3.59(\mathrm{~s}, 3 \mathrm{H}), 3.73(\mathrm{dd}, J=1.27 \mathrm{~Hz}, J=12.7$ $\mathrm{Hz}, 1 \mathrm{H}$ ), $3.89-4.01(\mathrm{~m}, 3 \mathrm{H}), 3.99(\mathrm{~d}, J=8.62 \mathrm{~Hz}, 1 \mathrm{H}), 4.06$ (d, $J=8.62 \mathrm{~Hz}, 1 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) $26.04,26.78,61.40$, $63.59,69.67,71.23,71.84,78.3,105.59$; IR $3408 \mathrm{~cm}^{-1}$; MS (CI) $m / e$ (relative intensity) $235(\mathrm{M}+1,3), 217$ (50), 177 (92), 159 (52), 145 (49), 127 (100), 99 (27), 74 (21). Anal. Calcd for $\mathrm{C}_{10} \mathrm{H}_{18} \mathrm{O}_{6}: \mathrm{C}, 51.28 ; \mathrm{H}, 7.74$. Found: C, $50.99 ; \mathrm{H}, 7.65$.

1,2-O-Isopropylidene-3- $O$-( $o$-methylbenzyl)- $\beta$-d-fructopyranose (9): $86 \%$ as a white solid; $[\alpha]_{\mathrm{D}}-79.5^{\circ}$ (c 0.146 , acetone); mp $92.9{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ) $\delta 1.40(\mathrm{~s}, 3 \mathrm{H}), 1.46$ (s, 3 H ), $2.35(\mathrm{~s}, 3 \mathrm{H}), 3.68(\mathrm{~d}, J=9.29 \mathrm{~Hz}, 1 \mathrm{H}), 3.74(\mathrm{dd}, J=$ $1.67 \mathrm{~Hz}, J=12.84 \mathrm{~Hz}, 1 \mathrm{H}), 3.89(\mathrm{~s}, 2 \mathrm{H}), 3.90-3.98(\mathrm{~m}, 3 \mathrm{H})$, $4.70(\mathrm{~d}, J=11.77 \mathrm{~Hz}, 1 \mathrm{H}), 4.82(\mathrm{~d}, J=11.77 \mathrm{~Hz}, 1 \mathrm{H}), 7.17-$ 7.33 (m, 4 H ); ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) 18.84, 26.21, 26.76, 63.58, $69.77,71.51,71.90,73.40,76.44,105.59,111.90,126.04$, $128.21,128.75,130.44,135.94,136.45$; IR $3402 \mathrm{~cm}^{-1}$; MS (EI) $m / e$ (relative intensity) $324\left(\mathrm{M}^{+}, 5\right), 177$ (11), 117 (20), 105 (100), 59 (15). Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{24} \mathrm{O}_{6}: \mathrm{C}, 62.95 ; \mathrm{H}, 7.46$. Found: C, 63.09; H, 7.55.

1,2-O-Isopropylidene-3-O-( $m$-methylbenzyl)- $\beta$-d-fructopyranose (10): $88 \%$ as a white solid; $[\alpha]_{\mathrm{D}}-116.5^{\circ}$ (c 0.103, acetone); mp $73.3^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.42(\mathrm{~s}, 3 \mathrm{H}), 1.47$ (s, $3 \mathrm{H}), 2.33(\mathrm{~s}, 3 \mathrm{H}), 2.36(\mathrm{~s}, 1 \mathrm{H}), 2.45(\mathrm{~s}, 1 \mathrm{H}), 3.65(\mathrm{~d}, J=9.05$ $\mathrm{Hz}, 1 \mathrm{H}), 3.74(\mathrm{dd}, J=12.86 \mathrm{~Hz}, J=1.60 \mathrm{~Hz}, 1 \mathrm{H}), 3.90-4.06$ $(\mathrm{m}, 5 \mathrm{H}), 4.70(\mathrm{~s}, 2 \mathrm{H}), 7.16(\mathrm{~m}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 21.34$, $26.13,26.84,63.51,69.63,71.22,71.80,75.40,76.52,105.57$, $111.93,124.99,128.59,128.71,128.85,137.79$; IR $3391 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $324\left(\mathrm{M}^{+}, 3\right), 309(2), 177(16)$, 146 (9), 235 (2), 190 (6), 117 (24). Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{24} \mathrm{O}_{6}$ : C, $62.95 ; \mathrm{H}, 7.46$. Found: C, $63.22 ; \mathrm{H}, 7.64$.

1,2-O-Isopropylidene-3- $O$-( $p$-methylbenzyl) $\beta$ - D -fructopyranose (11): $93 \%$ as a white solid; ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ) $\delta 1.41$ (s, 3 H ), 1.47 (s, 3 H ), 2.30 (b, 2 H ), 2.32 ( $\mathrm{s}, 3 \mathrm{H}$ ), 3.64 (d, $J=$ $8.87 \mathrm{~Hz}, 1 \mathrm{H}), 3.73$ (dd, $J=12.9 \mathrm{~Hz}, J=1.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.89-$ $4.04(\mathrm{~m}, 5 \mathrm{H}), 4.67(\mathrm{~d}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.71(\mathrm{~d}, J=12.0 \mathrm{~Hz}$, $1 \mathrm{H}), 7.13-7.25$ (m, 4 H ).

1,2-O-Isopropylidene-3- $O$-( $p$-methoxybenzyl)- $\beta$-d-fructopyranose (12): $85 \%$ as a white solid; ${ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) \delta$

Table 5. Effect of Alkyl Chain of $\beta$-D-Fructopyranose Auxiliary in Cyclopropanation

| starting material |  | cyclopropanation |  |  |  | hydrolysis-reduction |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{R}_{1}$ in 50 | no. | solvent | temp ( ${ }^{\circ} \mathrm{C}$ ) | time (h) | yield (\%) ${ }^{\text {a }}$ | yield (\%) ${ }^{a}$ | ee (49: \%) ${ }^{\text {b }}$ |
| Bn | 28A | $\left(\mathrm{ClCH}_{2}\right)_{2}$ | $-20 \rightarrow 0$ | 24 | 85 | 88 | 71 |
| Bn | 28A | $\left(\mathrm{ClCH}_{2}\right)_{2}$ | $-30 \rightarrow-20$ | 48 | $54^{\text {c }}$ | 86 | 83 |
| $\mathrm{CH}_{3}$ | 29A | $\left(\mathrm{ClCH}_{2}\right)_{2}$ | $-20 \rightarrow 0$ | 24 | 88 | 88 | 46 |
| $o-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | 30A | $\left(\mathrm{ClCH}_{2}\right)_{2}$ | $-25 \rightarrow 0$ | 22 | $72^{c}$ | 86 | 68 |
| $m-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | 31A | $\left(\mathrm{ClCH}_{2}\right)_{2}$ | $-25 \rightarrow 0$ | 22 | 87 | 89 | 72 |
| $p-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | 32A | toluene | $-20 \rightarrow 0$ | 22 | 86 | 86 | 75 |
| $p-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | 32A | $\left(\mathrm{ClCH}_{2}\right)_{2}$ | $-30 \rightarrow-10$ | 48 | $43^{c}$ | 90 | 82 |
| $p-\mathrm{MeOC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | 33A | $\left(\mathrm{ClCH}_{2}\right)_{2}$ | $-20 \rightarrow 0$ | 20 | 87 | 85 | 66 |
| $p$-t- $\mathrm{BuC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | 34A | $\left(\mathrm{ClCH}_{2}\right)_{2}$ | $-20 \rightarrow 0$ | 22 | 89 | 91 | 69 |
| $p-\mathrm{PhC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | 35A | $\left(\mathrm{ClCH}_{2}\right)_{2}$ | $-30 \rightarrow 0$ | 24 | 85 | 93 | 77 |
| $2-\mathrm{NpCH}_{2}$ | 36A | $\left(\mathrm{ClCH}_{2}\right)_{2}$ | $-30 \rightarrow 0$ | 24 | 85 | 90 | 64 |

${ }^{a}$ Isolated yield unless noted otherwise. ${ }^{b}$ The enantiomeric excess was determined by HPLC analysis (Daicel OJ, $10 \% i$-PrOH in hexane). ${ }^{c}$ Based on ${ }^{1} \mathrm{H}$ NMR.

$1.43(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 3 \mathrm{H}), 2.50(\mathrm{~b}, 2 \mathrm{H}), 3.67(\mathrm{~d}, J=8.99 \mathrm{~Hz}$, $1 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}), 3.69-4.05(\mathrm{~m}, 6 \mathrm{H}), 4.67(\mathrm{~d}, J=12.0 \mathrm{~Hz}$, $1 \mathrm{H}), 4.72(\mathrm{~d}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.89(\mathrm{~d}, J=8.64 \mathrm{~Hz}, J=8.66$ $\mathrm{Hz}, 2 \mathrm{H}$ ), 7.29 (d, 2 H ).

1,2- $O$-Isopropylidene-3- $O$-(p-tert-butylbenzyl)- $\beta$-D-fructopyranose (13): $93 \%$ as a white solid; $[\alpha]_{D}-86.9^{\circ}$ (c 0.107, acetone); mp $92{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.29(\mathrm{~s}, 9 \mathrm{H}), 1.42$ (s, $3 \mathrm{H}), 1.47(\mathrm{~s}, 3 \mathrm{H}), 3.67(\mathrm{~d}, J=9.18 \mathrm{~Hz}, 1 \mathrm{H}), 3.75(\mathrm{dd}, J=$ $12.3 \mathrm{~Hz}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.86-4.16(\mathrm{~m}, 5 \mathrm{H}), 4.72(\mathrm{~d}, J=11$ $\mathrm{Hz}, 1 \mathrm{H}), 4.75(\mathrm{~d}, J=11 \mathrm{~Hz}, 1 \mathrm{H}), 7.25(\mathrm{~d}, J=9.63 \mathrm{~Hz}, J=$ $8.39 \mathrm{~Hz}, 2 \mathrm{H}), 7.36(\mathrm{~d}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) 26.18,26.82$, $32.31,63.51,69.63,71.23,71.80,75.13,77.45,113.81,125.59$, 127.81; IR $3370 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $366\left(\mathrm{M}^{+}\right)$, 351 (1), 219 (5), 147 (100), 132 (14), 117 (38), 103 (76), 59 (11). Anal. Caled for $\mathrm{C}_{20} \mathrm{H}_{30} \mathrm{O}_{6}$ : $\mathrm{C}, 65.55 ; \mathrm{H}, 8.25$. Found: $\mathrm{C}, 65.17$; H, 8.51.

1,2-O-Isopropylidene-3- $O$ - $\boldsymbol{p}$-phenylbenzyl)- $\boldsymbol{\beta}$-d-fructopyranose (14): $85 \%$ as a white solid; $[\alpha]_{D}-100^{\circ}$ (c 0.143 , acetone); mp $155.2^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.44(\mathrm{~s}, 3 \mathrm{H}), 1.48$ $(\mathrm{s}, 3 \mathrm{H}), 2.20(\mathrm{~b}, 2 \mathrm{H}), 3.69(\mathrm{~d}, J=9.35 \mathrm{~Hz}, 1 \mathrm{H}), 3.76(\mathrm{dd}, J=$ $1.87 \mathrm{~Hz}, J=12.85 \mathrm{~Hz}, 1 \mathrm{H}), 3.93-4.10(\mathrm{~m}, 5 \mathrm{H}), 4.80(\mathrm{~s}, 2 \mathrm{H})$, 7.32-7.60(m, 9 H$) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 26.19,26.79,63.58$, $69.72,71.33,71.86,74.99,76.56,105.57,111.96,127.07$, $127.34,128.34,128.77,136.87,140.64,140.97$; IR $3369 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $386\left(\mathbf{M}^{+}\right), 371$ (25), 300 (20), 239 (17), 167 (100), 117 (11), 103 (46), 59 (33). Anal. Calcd for $\mathrm{C}_{22} \mathrm{H}_{26} \mathrm{O}_{6}$ : $\mathrm{C}, 68.38 ; \mathrm{H}, 6.78$. Found: $\mathrm{C}, 68.68 ; \mathrm{H}, 6.87$.

1,2-O-Isopropylidene-3- $O$-(2-naphthylmethyl)- $\boldsymbol{\beta}$-D-fructopyranose (15): $91 \%$ as a white solid; $[\alpha]_{D}-94.4^{\circ}$ (c 0.108, acetone); mp $128.3^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.45(\mathrm{~s}, 3 \mathrm{H}), 1.48$ $(\mathrm{s}, 3 \mathrm{H}), 2.15(\mathrm{~b}, 2 \mathrm{H}), 3.70(\mathrm{~d}, J=9.43 \mathrm{~Hz}, 1 \mathrm{H}), 3.75(\mathrm{dd}, J=$ $12.87 \mathrm{~Hz}, J=1.75 \mathrm{~Hz}, 1 \mathrm{H}), 3.93-4.10(\mathrm{~m}, 5 \mathrm{H}), 4.88(\mathrm{~d}, J=$ $12.03 \mathrm{~Hz}, 1 \mathrm{H}), 4.94(\mathrm{~d}, J=12.03 \mathrm{~Hz}, 1 \mathrm{H}), 7.44-7.45(\mathrm{~m}, 3$ $\mathrm{H}), 7.79-7.85(\mathrm{~m}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 26.19,26.81,63.56$, $69.73,71.38,71.88,75.42,76.59,105.58,111.98,125.69$, $126.12,126.29,126.76,127.72,127.94,128.51$; IR $3416 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $360\left(\mathrm{M}^{+}\right), 213$ (1), 167 (5), 141 (100), 103 (37), 59 (42), 43 (44). Anal. Calcd for $\mathrm{C}_{19} \mathrm{H}_{24} \mathrm{O}_{6}: \mathrm{C}$, 66.65; H, 6.71. Found: C, 66.96; H, 6.84 .

Preparation of $1,2: 4,5-\mathrm{Di}-\mathrm{O}$-isopropylidene- $\boldsymbol{\beta}$-D-psicopyranose (16). Pyridinium chlorochromate ( $9.7 \mathrm{~g}, 45$ mmol ) was added to 1,$2 ; 3,4$-di- $O$-isopropylidene- $\beta$-d-fructopyranose $(4,7.8 \mathrm{~g}, 30 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(80 \mathrm{~mL})$. The reaction mixture was well stirred for 3 h . When the oxidation was complete, the reaction mixture was diluted with diethyl ether and filtered through Florisil. Removal of the solvent gave 1,2:

4,5-di- $O$-isopropylidene- $\beta$-D-erythro-2,3-hexadiulopyranose (5 $\mathrm{g}, 65 \%$ ) as a colorless oil. To the resulting ketone ( $4.3 \mathrm{~g}, 16.6$ mmol ) in THF ( 150 mL ) was added lithium aluminum hydride $(1.27 \mathrm{~g}, 33.3 \mathrm{mmol})$ under nitrogen at $-78^{\circ} \mathrm{C}$. The reaction mixture was stirred for 1 h at $-78^{\circ} \mathrm{C}$. Workup was executed by the $\mathrm{X}-2 \mathrm{X}-\mathrm{X}$ rule. That is, 1.27 mL of water was added dropwise followed by addition of 2.54 mL of $15 \% \mathrm{NaOH}$ solution and finally 1.27 mL of water. The solution was filtered and concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel with $2: 1$ ether-hexane to give $1,2: 4,5$-di- $O$-isopropylidene- $\beta$-D-psicopyranose ( $16,4.03 \mathrm{~g}, 93 \%$ ) as a white solid: $[\alpha]_{\mathrm{D}}-110.9^{\circ}(c$ 0.238 , acetone); $\operatorname{mp} 65.8{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.36(\mathrm{~s}, 3 \mathrm{H})$, $1.38(\mathrm{~s}, 3 \mathrm{H}), 1.47(\mathrm{~s}, 3 \mathrm{H}), 1.53(\mathrm{~s}, 3 \mathrm{H}), 3.72(\mathrm{~d}, J=4.00 \mathrm{~Hz}$, $1 \mathrm{H}), 3.98(\mathrm{~m}, 2 \mathrm{H}), 4.02(\mathrm{~d}, J=9.29 \mathrm{~Hz}, 1 \mathrm{H}), 4.21(\mathrm{~m}, 1 \mathrm{H})$, 4.23 (d, $J=9.30 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.41 (dd, $J=3.99 \mathrm{~Hz}, J=6.73 \mathrm{~Hz}$, $1 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) 25.05,25.99,26.10,26.46,61.17,68.73$, $71.80,72.08,72.99,104.86,109.38,111.07 ;$ IR $3470 \mathrm{~cm}^{-1}$; MS (CI) m/e (́relative intensity) $261(\mathrm{M}+1,8), 245(10), 203(100)$, 185 (26), 145 (21), 127 (38), 99 (13), 85 (17), 69 (12). Anal. Calcd for $\mathrm{C}_{12} \mathrm{H}_{20} \mathrm{O}_{6}: \mathrm{C}, 55.37 ; \mathrm{H}, 7.74$. Found: C, $55.50 ; \mathrm{H}$, 7.86.

Preparation of $1,2-O$-Isopropylidene-3- $O$ - $(p$-phenyl-benzyl)- $\boldsymbol{\beta}$-D-psicopyranose (17). 1,2:3,4-Di- $O$-isopropylidene-$\beta$-D-psicopyranose ( $16,2.95 \mathrm{~g}, 11.3 \mathrm{mmol}$ ) was treated with $80 \%$ sodium hydride ( $408 \mathrm{mg}, 13.6 \mathrm{mmol}$ ) and $p$-phenylbenzyl chloride ( $2.53 \mathrm{~g}, 12.5 \mathrm{mmol}$ ) in $N, N$-dimethylformamide ( 35 mL ), followed by treatment of the crude product with $80 \%$ acetic acid in $\mathrm{H}_{2} \mathrm{O}(40 \mathrm{~mL})$. After evaporation of the solvent, the residue was purified by flash chromatography on silica gel with $2: 1$ hexane-acetone to give the benzylated psicopyranose 17 ( $3.46 \mathrm{~g}, 79 \%$ ) as a white solid: $[\alpha]_{\mathrm{D}}-80.9^{\circ}(c 0.216$, acetone); $\operatorname{mp} 117.8^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.35(\mathrm{~s}, 3 \mathrm{H}), 1.47(\mathrm{~s}, 3 \mathrm{H})$, $3.70 \sim 4.0(\mathrm{~m}, 6 \mathrm{H}), 4.08(\mathrm{~d}, J=9.22 \mathrm{~Hz}, 1 \mathrm{H}), 4.72(\mathrm{~d}, J=$ $11.25 \mathrm{~Hz}, 1 \mathrm{H}), 4.90(\mathrm{~d}, J=11.25 \mathrm{~Hz}, 1 \mathrm{H}), 7.33-7.60(\mathrm{~m}, 9$ $\mathrm{H})$; ${ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) 26.32,26.72,65.35,67.38,69.28,73.41$, $75.88,81.27,104.80,112.14,127.11,127.41,127.46,128.51$, $128.80,136.06,140.30,141.10$; IR $3408 \mathrm{~cm}^{-1}$; MS (EI) $m / e$ (relative intensity) $386\left(\mathrm{M}^{+}, 5\right), 325(20), 282(15), 239$ (30), 167 (100), 117 (11), 103 (54), 59 (11). Anal. Calcd for $\mathrm{C}_{22} \mathrm{H}_{26} \mathrm{O}_{6}$ : C, 68.38; H, 6.78. Found: C, $68.51 ; \mathrm{H}, 6.88$.

Preparation of trans-5-Phenyl-2-penten-1-al Diethyl Acetal. Pyridinium chlorochromate ( $2.85 \mathrm{~g}, 13.2 \mathrm{mmol}$ ) was added to 3-phenylpropan-1-ol ( $18,1.2 \mathrm{~g}, 8.8 \mathrm{mmol}$ ) in dichloromethane ( 25 mL ). The mixture was well stirred for 2 h . When the oxidation was complete, the reaction mixture was diluted with diethyl ether and filtered through Florisil. Removal of the solvent gave the pure 3-phenylpropanal ( 968 $\mathrm{mg}, 81.9 \%$ ) as a colorless oil. Ethyl (triphenylphosphoranylidene) acetate ( $2.96 \mathrm{~g}, 8.5 \mathrm{mmol}$ ) was added to 3-phenylpropanal ( $950 \mathrm{mg}, 7.1 \mathrm{mmol}$ ) in dichloromethane $(20 \mathrm{~mL})$ at room temperature. The mixture was stirred for 1 h at the same temperature. The mixture was diluted with $\mathrm{H}_{2} \mathrm{O}$ and then extracted with ethyl acetate. The combined extracts were washed with water and brine, dried over $\mathrm{MgSO}_{4}$, and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel with $1: 3$ ether-hexane to give ethyl trans-5-phenyl-2-pentenoate ( $19,1.29 \mathrm{~g}, 89 \%$ ) as a

Table 6. Effect of Structure of Acetals on Cyclopropanation

| entry | starting material |  |  | cyclopropanation |  |  | hydrolysis-reduction |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{R}_{2}$ | $\mathrm{R}_{3}$ | no. | temp ( ${ }^{\circ} \mathrm{C}$ ) | time (h) | yield (\%) | yield (\%) | ee, \% (config) of 53 |
| 1 | Ph | H | 35A | $-30 \rightarrow 0$ | 24 | 85 | 93 | $77^{a}\left(1 R, 2 R^{e}\right)$ |
| 2 | $\mathrm{PhCH}_{2} \mathrm{CH}_{2}$ | H | 37A | $-30 \rightarrow-10$ | 48 | $45^{j}$ | 80 | $69^{b}(1 R, 2 R f)$ |
| 3 | $\mathrm{PhCH}_{2} \mathrm{CH}_{2}$ | H | 37A | $-30 \rightarrow 0$ | 35 | 87 | 88 | $52^{\text {b }}(1 R, 2 R t)$ |
| 4 | $\mathrm{BnOCH}_{2}$ | H | 38A | $-30 \rightarrow 0$ | 24 | 90 | 89 | $21^{\text {a }}\left(1 R, 2 R^{8}\right)$ |
| 5 | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | 39A | $-15 \rightarrow 0$ | 24 | 81 |  | $60^{c}\left(1 R^{h}\right)$ |
| 6 | $n-\mathrm{C}_{5} \mathrm{H}_{11}$ | H | 40A | $-30 \rightarrow-5$ | 35 | 69 | 82 | $81^{d}\left(1 R, 2 R^{i}\right)$ |
| 7 | $n-\mathrm{C}_{5} \mathrm{H}_{11}$ | H | 40A | $-15 \rightarrow 0$ | 48 | 90 | 85 | $64^{d}\left(1 R, 2 R^{i}\right)$ |

${ }^{a}$ HPLC analysis (Daicel OJ, $10 \% i$-PrOH in hexane). ${ }^{b}$ HPLC analysis (Daicel OD, $5 \% i$-PrOH in hexane). ${ }^{c}$ GC analysis of the corresponding amide with $(R)-(+)$ - $\alpha$-methylbenzylamine (Chiraldex B-PH GC column, 30 m , flow rate $1.01 \mathrm{~mL} / \mathrm{min}$, oven temp $150^{\circ} \mathrm{C}$ ). ${ }^{d}$ GC analysis of the corresponding ester with ( $R$ )-( + )-MTPA (Chiraldex G-TA GC column, 30 m , flow rate 1.00 mL , oven temp $140{ }^{\circ} \mathrm{C}$ ). ${ }^{e}[\alpha]_{\mathrm{D}}-70.3^{\circ}(c 0.26, \mathrm{EtOH})\left\{\mathrm{lit} .{ }^{11 \mathrm{c}}[\alpha]_{\mathrm{D}}-92^{\circ}(c 1.23, \mathrm{EtOH})\right\} .{ }^{f}[\alpha]_{\mathrm{D}}-17.1^{\circ}$ (c $1.5, \mathrm{CHCl}_{3}$ as $52 \%$ ee) $\left\{\mathrm{lit}.{ }^{17}[\alpha]_{\mathrm{D}}-24.6^{\circ}\left(c 1.13, \mathrm{CHCl}_{3}\right.\right.$ as $80 \%$ ee $)\} . s[\alpha]_{\mathrm{D}}-3.05^{\circ}$ (c $0.6, \mathrm{CHCl}_{3}$ as $21 \%$ ee) $\left\{\mathrm{lit} .{ }^{1 \mathrm{~b}}-6.0^{\circ} \text { ( } c 1.02, \mathrm{CHCl}_{3} \text { as } 35 \% \text { ee) }\right\}^{.}{ }^{h}$ The absolute configuration was determined by the comparison of its specific rotation with the literature value after transformation to the known ( $1 R$ )-2,2-dimethylcyclopropanecarboxylic acid from pure cyclopropanated acetal 52 by the following sequence: (i) $\mathrm{O}_{3} /$ ethyl acetate, (ii) $\mathrm{NaOH} / \mathrm{EtOH}$. $\left\{[\alpha]_{\mathrm{D}}-85^{\circ}\right.$ (c $0.1, \mathrm{CHCl}_{3}$ ) lit. ${ }^{14}$ ent-form $\left.[\alpha]_{\mathrm{D}}+142^{\circ}\left(c 1.01, \mathrm{CHCl}_{3}\right)\right]$. ${ }^{i}$ The absolute configuration was determined by the comparison of its specific rotation with the reported value after transformation to the known ( $1 R, 2 R$ )-methyl trans-2-pentylcyclopropanecarboxylate from pure 2-pentylcyclopropanecarboxaldehyde by the following reaction: (i) $\mathrm{MnO}_{2}, \mathrm{NaCN}, \mathrm{AcOH}, \mathrm{MeOH}\left\{[\alpha]_{\mathrm{D}}-49.2^{\circ}\right.$ ( ( $0.1, \mathrm{CHCl}_{3}$ as $64 \%$ ee), lit. ${ }^{15}$ ent-form $[\alpha]_{\mathrm{D}}+71^{\circ}$ (c $2.6, \mathrm{CHCl}_{3}$ as $92 \%$ ee) \}. ${ }^{j}$ The yield was based on ${ }^{1} \mathrm{H}$ NMR.

colorless oil: [ ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.26(\mathrm{t}, 3 \mathrm{H}), 2.50(\mathrm{~m}, 2 \mathrm{H})$, 2.76 (t, 2 H ), 4.16 (q, 2 H ), $5.82(\mathrm{~d}, J=15.68 \mathrm{~Hz}, 1 \mathrm{H}), 6.99$ (dt, 1 H ), 7.14-7.31 (m, 5 H$)$ ]. A solution of 1.0 M diisobutylaluminum hydride in hexane ( $35.3 \mathrm{~mL}, 35.3 \mathrm{mmol}$ ) was added to a stirred solution of ethyl trans-5-phenyl-2-pentenoate (19, $1.2 \mathrm{~g}, 5.88 \mathrm{mmol}$ ) in benzene ( 15 mL ) under nitrogen at $0^{\circ} \mathrm{C}$. After the solution was stirred overnight at room temperature, the reaction was stopped by addition of methanol ( 2.5 g ) in benzene ( 5 mL ) followed by water ( 0.15 g ). The aluminum salts were filtered, and the filtrate was concentrated under reduced pressure. The residue was purified by flash chromatography to give the corresponding alcohol ( $773 \mathrm{mg}, 81.1 \%$ ) as a colorless oil, to which a portion ( $740 \mathrm{mg}, 4.6 \mathrm{mmol}$ ) in dichloromethane ( 15 mL ) was added pyridinium chlorochromate ( $1.5 \mathrm{~g}, 6.9 \mathrm{mmol}$ ). The reaction mixture was well stirred for 1 h . When the oxidation was complete, the reaction mixture was diluted with diethyl ether and filtered through Florisil. The residue was purified by flash chromatography to give trans-5-phenyl-2-penten-1-al ( $\mathbf{2 0}, 587 \mathrm{mg}, 80 \%$ ) as a colorless oil. A mixture of the aldehyde $20(420 \mathrm{mg}, 2.63$ mmol ), triethyl orthoformate ( $0.53 \mathrm{~mL}, 3.15 \mathrm{mmol}$ ), and a catalytic amount of ammonium nitrate ( 21 mg ) in ethanol ( 26 mL ) was stirred at room temperature until consumption of most of the starting aldehyde was completed (by TLC). After the solvents were evaporated, the residue was worked up with ethyl acetate and aqueous sodium bicarbonate. The organic layer was separated and dried over $\mathrm{MgSO}_{4}$. Removal of the solvent left the corresponding diethyl acetal ( $584 \mathrm{mg}, 95 \%$ ), which was used for the next reaction without further purification: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.18(\mathrm{t}, 6 \mathrm{H}), 2.40(\mathrm{~m}, 2 \mathrm{H}), 2.71(\mathrm{t}, 2$ H), $3.54(\mathrm{~m}, 4 \mathrm{H}), 4.80(\mathrm{~d}, J=5.51 \mathrm{~Hz}, 1 \mathrm{H}), 5.53(\mathrm{dd}, J=5.62$ $\mathrm{Hz}, J=15.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.81(\mathrm{dt}, 1 \mathrm{H}), 7.13-7.30(\mathrm{~m}, 5 \mathrm{H})$.

Preparation of trans-4-(Benzyloxy)-2-buten-1-al Diethyl Acetal. To a solution of trans-2-butene-1,4-diol (21, 2.5 $\mathrm{g}, 28.4 \mathrm{mmol}$ ) in DMF ( 40 mL ) was added $80 \%$ sodium hydride ( $426 \mathrm{mg}, 14.2 \mathrm{mmol}$ ) at room temperature. After 0.5 h of stirring, benzyl bromide ( $1.7 \mathrm{~mL}, 14.2 \mathrm{mmol}$ ) was added to the above solution, which was stirred for 1 h at room temperature. The reaction mixture was diluted with water and then extracted with ethyl acetate. The combined extracts were washed with water and brine, dried over $\mathrm{MgSO}_{4}$, and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel with $1: 2$ ethyl acetatehexane to give the monobenzylated alcohol $22(1.9 \mathrm{~g}, 38 \%)$ as a colorless oil ${ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.65(\mathrm{~b}, \mathrm{OH}), 4.03(\mathrm{~d}, 2 \mathrm{H}, J$
$=4.1 \mathrm{~Hz}), 4.14(\mathrm{~d}, 2 \mathrm{H}, J=3.91 \mathrm{~Hz}), 4.52(\mathrm{~s}, 2 \mathrm{H}), 5.87(\mathrm{~m}, 2$ H), 7.31 ( $\mathrm{m}, 5 \mathrm{H}$ )]. Pyridinium chlorochromate ( $3.14 \mathrm{~g}, 14.6$ $\mathrm{mmol})$ was added to the monobenzylated alcohol $22(1.3 \mathrm{~g}, 7.3$ mmol ) in dichloromethane ( 15 mL ). The mixture was well stirred for 2 h . When the oxidation was complete, the reaction mixture was diluted diethyl ether and filtered through Florisil. After evaporation of the solvent, the residue was purified by flash chromatography on silica gel with 1:4 ethyl acetatehexane to give trans-4-(benzyloxy)-2-buten-1-al (23, 1.03 g , $80 \%$ ) as a colorless oil [ ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ) $\delta 4.27$ (dd, $J=1.93$ $\mathrm{Hz}, J=4.0 \mathrm{~Hz}, 2 \mathrm{H}$ ), 4.58 ( $\mathrm{s}, 2 \mathrm{H}$ ), 6.42 (ddt, $J=15.7 \mathrm{~Hz}, J$ $=7.86 \mathrm{~Hz}, 1 \mathrm{H}$, $6.84(\mathrm{dt}, J=15.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.33(\mathrm{~m}, 5 \mathrm{H}), 9.57$ (d, $J=7.96 \mathrm{~Hz}, 1 \mathrm{H}$ ); ${ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) 68.58,73.02,127.67$, 127.97, 128.53, 131.86, 137.40, 152.93, 173.8]. A mixture of the enal 23 ( $490 \mathrm{mg}, 2.78 \mathrm{mmol}$ ), triethyl orthoformate ( 0.56 $\mathrm{mL}, 3.34 \mathrm{mmol}$ ), and a catalytic amount of ammonium nitrate ( 23 mg ) in ethanol ( 30 mL ) was stirred at room temperature until the consumption of the starting aldehyde was confirmed by TLC. After the solvents were evaporated, the residue was worked up with ethyl acetate and aqueous sodium bicarbonate. The organic layer was separated and dried over $\mathrm{MgSO}_{4}$. Removal of the solvent left the corresponding diethyl acetal ( $656 \mathrm{mg}, 94 \%$ ), which was used for next reaction without further purification: ${ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.20(\mathrm{t}, 6 \mathrm{H}), 3.59(\mathrm{~m}$, 4 H ), $4.04(\mathrm{~d}, J=5.24 \mathrm{~Hz}, 2 \mathrm{H}), 4.51(\mathrm{~s}, 2 \mathrm{H}), 4.90(\mathrm{~d}, J=4.85$ $\mathrm{Hz}, 1 \mathrm{H}), 5.74$ (dd, $J=4.74 \mathrm{~Hz}, J=15.53 \mathrm{~Hz}, 1 \mathrm{H}), 5.95$ (dt, $J=15.53 \mathrm{~Hz}, 1 \mathrm{H}), 7.31(\mathrm{~m}, 5 \mathrm{H})$.

Preparation of 6-Phenyl-2-hexyn-1-al Diethyl Acetal (26). A solution of $1.6 \mathrm{M} n$-butyllithium in hexane ( 14.6 mL , $23.3 \mathrm{mmol})$ was added to a stirred solution of 3-( tert-butyldiphenylsilyl)oxy)propyne ( $6.25 \mathrm{~g}, 21.25 \mathrm{mmol}$ ) and hexamethylphosphoric triamide ( $11.1 \mathrm{~mL}, 63.8 \mathrm{mmol}$ ) in THF ( 50 mL ) under nitrogen at $-78^{\circ} \mathrm{C}$. The temperature was progressively raised to $-15{ }^{\circ} \mathrm{C}$ and maintained for 40 min . 1-Bromo-3phenylpropane ( $\mathbf{2 4}, 4.44 \mathrm{~g}, 22.3 \mathrm{mmol}$ ) was added dropwise at $-78^{\circ} \mathrm{C}$, and the resulting solution was stirred for 5 h at room temperature. The reaction mixture was washed with water and then extracted with ethyl acetate. The combined extracts were washed with water and brine, dried over $\mathrm{MgSO}_{4}$, and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel with $1: 4$ chloroformhexane to give 1-((tert-butyldiphenylsilyl)oxy)-6-phenyl-2hexyne ( $7.37 \mathrm{~g}, 80 \%$ ) as a colorless oil [ ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.2$ $(\mathrm{s}, 9 \mathrm{H}), 1.78(\mathrm{~m}, 2 \mathrm{H}), 2.19(\mathrm{~m}, 2 \mathrm{H}), 2.65(\mathrm{t}, 2 \mathrm{H}), 4.35(\mathrm{t}, 2$ $\mathrm{H}), 7.1-7.8(\mathrm{~m}, 15 \mathrm{H})$. To a solution of 1-( (tert-butyldiphe-nylsilyl)oxy)-6-phenyl-2-hexyne ( $7 \mathrm{~g}, 17 \mathrm{mmol}$ ) in THF ( 50 mL ) was added tetrabutylammonium fluoride ( $6.7 \mathrm{~g}, 25.5 \mathrm{mmol}$ ) at room temperature. After 1 h , the mixture was concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel to give 6-phenyl-2-hexyn-1-ol (25, $2.6 \mathrm{~g}, 88 \%$ ) as a colorless oil [ ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.61(\mathrm{~b}, \mathrm{OH})$, $1.83(\mathrm{~m}, 2 \mathrm{H}), 2.21(\mathrm{~m}, 2 \mathrm{H}), 2.70(\mathrm{t}, 2 \mathrm{H}), 4.24(\mathrm{~s}, 2 \mathrm{H}), 7.24$ ( $\mathrm{m}, 5 \mathrm{H}$ ); IR $3346 \mathrm{~cm}^{-1}$ ]. Pyridinium chlorochromate ( $2 \mathrm{~g}, 9.48$

Table 7. Simmons-Smith Reaction of Exo Acetals

| starting material |  |  | cyclopropanation |  |  | hydrolysis-reduction |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{R}_{1}$ | $\mathrm{R}_{2}$ | no. | temp ( ${ }^{\circ} \mathrm{C}$ ) | time ( h ) | yield (\%) | yield (\%) | ee, $\%^{a}$ (config) of 55 |
| Bn | Ph | $28 B$ | $-30 \rightarrow 0$ | 24 | 82 | 72 | $33^{\text {b }}$ ( $1 S, 2 S^{e}$ ) |
| $p$ - $\mathrm{PhC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | $\mathrm{PhCH}_{2} \mathrm{CH}_{2}$ | 37B | $-30 \rightarrow-10$ | 24 | 88 | 85 | $45^{c}\left(1 S, 2 S^{\prime}\right)$ |
| $p-\mathrm{PhC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | $n-\mathrm{C}_{5} \mathrm{H}_{11}$ | 40B | $-30 \rightarrow-5$ | 35 | 84 | 84 | $54^{d}\left(1 S, 2 S^{g}\right)$ |

${ }^{a}$ The enantiomeric excess was determined by HPLC analysis. ${ }^{b}$ Daicel OJ, $10 \% i$-PrOH in hexane. ${ }^{c}$ Daicel OD, $5 \% i$-PrOH in hexane. ${ }^{d} \mathrm{GC}$ analysis of the corresponding ester with $(R)-(+)-\mathrm{MTPA}$ (Chiraldex G-TA GC column 30 m , flow rate 1.00 mL , oven temp $140{ }^{\circ} \mathrm{C}$ ). ${ }^{e}[\alpha]_{\mathrm{D}}+37.9^{\circ}(c 0.14, \mathrm{EtOH})\left\{\mathrm{lit} .{ }^{11}\right.$ ent-form $\left.[\alpha]_{\mathrm{D}}-92^{\circ}(c 1.23, \mathrm{EtOH})\right\} . f[\alpha]_{\mathrm{D}}+14.2^{\circ}\left(c 0.7, \mathrm{CHCl}_{3}\right.$ as $45 \%$ ee $)\left\{\right.$ lit. ${ }^{17}$ ent-form $[\alpha]_{\mathrm{D}}-24.6^{\circ}$ ( $c 1.13, \mathrm{CHCl}_{3}$ as $80 \%$ ee) \}. $\varepsilon$ The absolute configuration was determined by the comparison of its specific rotation with a literature value after transformation to ( $1 S, 2 S$ )-methyl trans-2-pentylcyclopropanecarboxylate: $\left.[\alpha]_{\mathrm{D}}+40.2^{\circ}(c) 1.9, \mathrm{CHCl}_{3}\right)\left\{\right.$ [itt. ${ }^{15}[\alpha]_{\mathrm{D}}+71^{\circ}$ (c $2.6, \mathrm{CHCl}_{3}$ as $92 \%$ ee) $\}$.

mmol ) was added to the alcohol $25(1.1 \mathrm{~g}, 6.32 \mathrm{mmol})$ in dichloromethane ( 15 mL ). The mixture was well stirred for 2 h. When the oxidation was complete, the reaction mixture was diluted with diethyl ether and filtered through Florisil. After evaporation of the solvent, the residue was purified by flash chromatography on silica gel with $1: 2$ ether-hexane to give 6 -phenyl-2-hexyn-1-al ( $931 \mathrm{mg}, 86 \%$ ) as a colorless oil [ ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.87(\mathrm{~m}, 2 \mathrm{H}), 2.39(\mathrm{~m}, 2 \mathrm{H}), 2.73(\mathrm{t}, 3 \mathrm{H}), 7.25(\mathrm{~m}$, $5 \mathrm{H}), 9.17(\mathrm{~s}, 1 \mathrm{H})$; IR 2201, $1666 \mathrm{~cm}^{-1}$ ]. A mixture of 6 -phenyl-2-hexyn-1-al ( $800 \mathrm{mg}, 4.65 \mathrm{mmol}$ ), triethyl orthoformate ( 0.93 $\mathrm{mL}, 5.58 \mathrm{mmol}$ ), and a catalytic amount of ammonium nitrate ( 37 mg ) in ethanol ( 50 mL ) was stirred at room temperature until the consumption of the starting aldehyde was complete (TLC). After the solvent was evaporated, the residue was worked up with ethyl acetate and aqueous sodium bicarbonate. The organic layer was separated and dried over $\mathrm{MgSO}_{4}$. Removal of the solvent left the corresponding diethyl acetal $26(1.03 \mathrm{~g}, 90 \%)$, which was used for next reaction without further purification: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.21(\mathrm{~m}, 6 \mathrm{H}), 1.84$ ( $\mathrm{m}, 2 \mathrm{H}$ ), $2.25(\mathrm{~m}, 2 \mathrm{H}), 2.72(\mathrm{~m}, 2 \mathrm{H}), 3.57(\mathrm{~m}, 4 \mathrm{H}), 5.25(\mathrm{t}, 1$ H), $7.25(\mathrm{~m}, 5 \mathrm{H})$.

General Procedure for Acetalization of $\alpha, \beta$-Unsaturated Aldehyde Diethyl Acetal with Sugar Diol: Preparation of 3-O-Benzyl-1,2-O-isopropylidene-4,5-O-[(1'R) (and (1'S))-trans-3'-phenyl-2'-propen-1'-yll- $\beta$-d-fructopyranose (28A and 28B). A mixture of the trans-cinnamaldehyde ( $10 \mathrm{~mL}, 79.3 \mathrm{mmol}$ ), triethyl orthoformate ( $15.8 \mathrm{~mL}, 95.2$ mmol ), and a catalytic amount of ammonium nitrate ( 7.9 $\mathrm{mmol}, 635 \mathrm{mg}$ ) in ethanol ( 80 mL ) was stirred at room temperature until the consumption of the starting aldehyde was confirmed by TLC. The solvent was evaporated, and the residue was worked up with ethyl acetate and aqueous sodium bicarbonate. The organic layer was separated and dried over anhydrous sodium sulfate. Removal of the solvent left the corresponding diethyl acetal, which was used for the next reaction without further purification. A mixture of the transcinnamaldehyde diethyl acetal ( $725 \mathrm{mg}, 3.5 \mathrm{mmol}$ ), a catalytic amount of pyridinium tosylate ( 25 mg ), and $3-O$-benzyl-1,2-$O$-isopropylidene- $\beta$-D-fructopyranose ( $\mathbf{7}, 1.2 \mathrm{~g}, 3.87 \mathrm{mmol}$ ) in 30 mL of benzene was heated for 24 h . After the solvent was evaporated, the residue was purified by flash chromatography (ether:hexane:chloroform $=1: 6: 2$ ) on silica gel to give white solids of 3-O-benzyl-1,2-O-isopropylidene-4,5-O-[(1'R)-trans-3'-phenyl-2'-propen-1'-yl]- $\beta$-D-fructopyranose ( $\mathbf{2 8 A}, 888 \mathrm{mg}, 58 \%$ ) and of 3 -O-benzyl-1,2-O-isopropylidene-4,5-O-[(1'S)-trans-3'-phenyl-2'-propen-1'-yll- $\beta$-D-fructopyranose (28B, $492 \mathrm{mg}, 32 \%$ ).

3-O-Benzyl-1,2-O-isopropylidene-4,5-O-[(1'R)-trans-3'-phenyl-2'-propen-1'-yl]- $\beta$-D-fructopyranose (28A): $[\alpha]_{D}$ $-126.7^{\circ}$ (c 0.105 , acetone); mp $97.6^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.42$ $(\mathrm{s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 3 \mathrm{H}), 3.49(\mathrm{~d}, J=7.12 \mathrm{~Hz}, 1 \mathrm{H}), 3.86(\mathrm{~d}, J=$ $8.45 \mathrm{~Hz}, 1 \mathrm{H}$ ), $4.07-4.21(\mathrm{~m}, 4 \mathrm{H}), 4.44(\mathrm{dd}, J=5.77 \mathrm{~Hz}, J=$ $7.01 \mathrm{~Hz} 1 \mathrm{H}), 4.68(\mathrm{~d}, J=12.04 \mathrm{~Hz}, 1 \mathrm{H}), 4.93(\mathrm{~d}, J=12.04$ $\mathrm{Hz}, 1 \mathrm{H}) 5.52(\mathrm{~d}, J=6.62 \mathrm{~Hz}, 1 \mathrm{H}), 6.12(\mathrm{dd}, J=6.57 \mathrm{~Hz}, J$ $=15.98 \mathrm{~Hz}, 1 \mathrm{H}), 6.79(\mathrm{~d}, J=15.99 \mathrm{~Hz}, 1 \mathrm{H}), 7.24-7.44(\mathrm{~m}$, 10 H ); ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) 26.02, 26.90, 59.80, 71.80, 72.66, $75.88,76.36,77.37,104.38,112.29,125.65,127.02,127.65$, $128.00,128.28,128.60,135.53,135.64$; IR $3030,1658 \mathrm{~cm}^{-1}$; MS (CI) m/e (relative intensity) $425(\mathrm{M}+1,34), 367$ (22), 349 (9), 303 (9), 259 (6), 213 (9), 185 (12), 133 (19), 117 (28), 91 (100), 69 (12). Anal. Calcd for $\mathrm{C}_{25} \mathrm{H}_{28} \mathrm{O}_{6}: \mathrm{C}, 70.74 ; \mathrm{H}, 6.65$. Found: C, 70.69; H, 6.59.

3-O-Benzyl-1,2-O-isopropylidene-4,5-O-[(1'S)-trans-3'-phenyl-2'-propen- $1^{\prime}$-yl]- $\boldsymbol{\beta}$-D-fructopyranose (28B): $[\alpha]_{\mathrm{D}}$ $-65.9^{\circ}$ (c 0.135 , acetone); $\mathrm{mp} 87.2^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $\left(\mathrm{CDCl}_{3}\right) \delta 1.42$ ( $\mathrm{s}, 3 \mathrm{H}$ ), 1.49 (s, 3 H ), 3.52 (d, $J=7.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), $3.88(\mathrm{~d}, J=$ $8.38 \mathrm{~Hz}, 1 \mathrm{H}), 4.05-4.19(\mathrm{~m}, 4 \mathrm{H}), 4.58(\mathrm{dd}, J=5.31 \mathrm{~Hz}, J=$ $7.42 \mathrm{~Hz}, 1 \mathrm{H}), 4.71(\mathrm{~d}, J=11.85 \mathrm{~Hz}, 1 \mathrm{H}), 4.97(\mathrm{~d}, J=11.85$ $\mathrm{Hz}, 1 \mathrm{H}), 5.69(\mathrm{~d}, J=6.09 \mathrm{~Hz}, 1 \mathrm{H}), 6.14$ (dd, $J=15.89 \mathrm{~Hz}, J$ $=6.12 \mathrm{~Hz}, 1 \mathrm{H}), 6.73(\mathrm{~d}, J=15.99 \mathrm{~Hz}, 1 \mathrm{H}), 7.24-7.43(\mathrm{~m}, 10$ H ); ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) 26.03, 26.91, 60.16, 71.78, 72.79, 73.08, $73.75,78.53,103.13,104.36,112.41,125.59,126.95,127.77$, 128.01, 128.38, $128.44,128.60,134.34,135.64,137.88$; IR 3032 , 2993, 2935, $1655 \mathrm{~cm}^{-1}$; MS (CI) m/e (relative intensity) 425 ( $\mathrm{M}+1$ ), 367 (19), 349 (6), 321 (6), 259 (5), 213 (7), 185 (10), 173 (8), 133 (26), 117 (29), 91 (100), 69 (13). Anal. Calcd for $\mathrm{C}_{25} \mathrm{H}_{28} \mathrm{O}_{6}$ : C, 70.74; H, 6.65. Found: C, $70.50 ; \mathrm{H}, 6.74$.

Using the procedure given above, the following compounds were prepared.

1,2-O-Isopropylidene-3-O-methyl-4,5-O-[(1'R)-trans-3'-phenyl-2'-propen- $1^{\prime}$-yll- $\beta$-d-fructopyranose (29A): $49 \%$ as a white solid; $[\alpha]_{\mathrm{D}}-85.1^{\circ}$ (c 0.047 , acetone); $\mathrm{mp} 111.9^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ) $\delta 1.41$ ( $\mathrm{s}, 3 \mathrm{H}$ ), $1.49(\mathrm{~s}, 3 \mathrm{H}), 3.28(\mathrm{~d}, J=7.26$ $\mathrm{Hz}, 1 \mathrm{H}), 3.57(\mathrm{~s}, 3 \mathrm{H}), 3.95(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.01-4.2(\mathrm{~m}$, 4 H ) 4.34 (dd, $J=5.47 \mathrm{~Hz}, J=7.15 \mathrm{~Hz}, 1 \mathrm{H}), 5.51(\mathrm{~d}, J=$ $6.31 \mathrm{~Hz}, 1 \mathrm{H}), 6.19(\mathrm{dd}, J=6.33 \mathrm{~Hz}, J=15.93 \mathrm{~Hz}, 1 \mathrm{H}), 6.79$ (d, $J=15.90 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) $25.99,26.83,59.89$, $60.05,71.90,75.97,77.09,79.29,104.21,104.30,125.57$, $126.98,128.54,128.62,135.59$; IR $2901,1662 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) 348 ( $\mathrm{M}^{+}, 3$ ), 333 (3), 232 (11), 205 (15), 159 (29), 131 (89), 115 (59), 104 (100), 87 (62), 84 (46), 59 (33), 43 (46). Anal. Calcd for $\mathrm{C}_{19} \mathrm{H}_{24} \mathrm{O}_{6}$ : C, $65.50 ; \mathrm{H}, 6.94$. Found: C, 65.76; H, 7.06.

1,2-O-Isopropylidene-3-O-(o-methylbenzyl)-4,5-O-[(1'R)-trans-3'-phenyl-2'-propen-1'-yll- $\beta$-D-fructopyranose (30A): $53 \%$ as a white solid; $[\alpha]_{D}-112.7^{\circ}$ (c 0.118 , acetone); mp 96.8 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.40(\mathrm{~s}, 3 \mathrm{H}), 1.48(\mathrm{~s}, 3 \mathrm{H}), 2.33(\mathrm{~s}, 3$ H), 3.51 (d, $J=7.22 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.82 (d, $J=8.38 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.93 (d, $J=8.38 \mathrm{~Hz}, 1 \mathrm{H}$ ), $4.05-4.21$ (m, 3 H ), 4.44 (dd, $J=5.63$ $\mathrm{Hz}, J=6.97 \mathrm{~Hz}, 1 \mathrm{H}), 4.65(\mathrm{~d}, J=12.06 \mathrm{~Hz}, 1 \mathrm{H}), 4.96(\mathrm{~d}, J$ $=12.06 \mathrm{~Hz}, 1 \mathrm{H}), 5.54(\mathrm{~d}, J=6.34 \mathrm{~Hz}, 1 \mathrm{H}), 6.20(\mathrm{dd}, J=6.38$ $\mathrm{Hz}, J=15.93 \mathrm{~Hz}, 1 \mathrm{H}), 6.82(\mathrm{~d}, J=15.90 \mathrm{~Hz}, 1 \mathrm{H}), 7.10-7.45$ (m, 9 H ); ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) 18.77, 26.08, 26.63, 59.86, 71.12, $71.96,76.03,76.08,77.46,104.27,104.38,112.26,125.59$, $125.64,126.99,127.89,128.57,128.64,129.01,130.23,135.53$, $135.83,136.79$; IR $3028,1658 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $438\left(\mathrm{M}^{+}, 2\right), 318(2), 229(16), 194(20), 171$ (44),

Table 8. Simmons-Smith Reactions of Cis Enal Acetals

| starting material |  |  | cyclopropanation |  |  | hydrol | oxidn | reduction |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{R}_{1}$ | $\mathrm{R}_{2}$ | no. | temp (Time, h) | yield ${ }^{\text {b }}$ (\%) | 56:57 ${ }^{\text {a }}$ | yield ${ }^{\text {b }}$ (\%) | no. (yield, \%) ${ }^{6}$ | no. | yield ${ }^{\text {b }}$ (\%) | ee (\%) |
| Bn | Ph | 45 | $-20 \rightarrow 0{ }^{\circ} \mathrm{C}$ (20) | 96 | 2:1 | 93 | 58 (55) ${ }^{\text {c }}$ | 60 | $89^{d}$ | $82^{e}$ |
|  |  |  |  |  |  |  | $59(28)^{\text {c }}$ | 61 | $90^{\prime}$ | $58{ }^{e}$ |
| p- $\mathrm{PhC}_{6} \mathrm{H}_{4} \mathrm{CH}_{2}$ | $\mathrm{Ph}\left(\mathrm{CH}_{2}\right)_{3}$ | 47 | $-30 \rightarrow 0{ }^{\circ} \mathrm{C}$ (24) | 89 | 56 only |  |  | 60 | $88^{8}$ | $80^{h}$ |

${ }^{a}$ The ratio was determined by ${ }^{1} \mathrm{H}$ NMR. ${ }^{b}$ Isolated yield. ${ }^{c} 58$ : $[\alpha]_{D}-29.7^{\circ}\left(c 0.175, \mathrm{CHCl}_{3}\right)\left\{\right.$ lit. ${ }^{13}$ ent-form $\left.\left.[\alpha]_{\mathrm{D}}+32.8^{\circ}(c) 1.99, \mathrm{CHCl}_{3}\right)\right\}$. 59: $[\alpha]_{\mathrm{D}}+173^{\circ}\left(c 0.185, \mathrm{CHCl}_{3}\right)\left\{\right.$ lit. $\left.{ }^{13}[\alpha]_{\mathrm{D}}+324.7^{\circ}\left(c \quad 1.24, \mathrm{CHCl}_{3}\right)\right\}$. ${ }^{d}$ Yield from the pure cis-methyl ester 58. ${ }^{e}$ HPLC analysis (Daicel OJ, $10 \%$ - PrOH in hexane). $f$ Yield from the pure trans-methyl ester 59.8 Yield for two steps (hydrolysis, reduction) from the isolated cis acetal of 56. ${ }^{h}$ HPLC analysis of the corresponding urethane with ( $R$ ) - $(+)$ - $\alpha$-methylbenzyl isocyanate (Daicel OD, $10 \% i-\mathrm{PrOH}$ in hexane). The absolute configuration of ( $1 R, 2 S$ )-cis-1-(hydroxymethyl)-2-(3'-phenylpropyl)-cyclopropane was determined by the comparison of its specific rotation with a literature value. $\left\{[\alpha]_{D}+18.2^{\circ}(c 0.77, \mathrm{EtOH})\right.$, lit. ${ }^{11 a}[\alpha]_{D}+19^{\circ}(c 0.7, \mathrm{EtOH}$ as $81 \%$ ee $\left.)\right\}$.


131 (31), 115 (26), 105 (100). Anal. Cacld for $\mathrm{C}_{26} \mathrm{H}_{30} \mathrm{O}_{6}$ : C, 71.21; H, 6.89. Found: C, 71.56; H, 7.01 .

1,2-O-Isopropylidene-3-O-(o-methylbenzyl)-4,5-O-[(1'S)-trans-3'-phenyl-1'-propen- $\left.1^{\prime} \cdot \mathrm{yl}\right]-\beta$-D-fructopyranose (30B): $35 \%$ as a white solid; [ $\alpha]_{\mathrm{D}}-47.90^{\circ}$ ( $c 0.192$, acetone); mp 109.2 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ) $\delta 1.43$ (s, 3 H ), 1.49 ( $\mathrm{s}, 3 \mathrm{H}$ ), 2.38 ( $\mathrm{s}, 3$ H), $3.56(\mathrm{~d}, J=7.25 \mathrm{~Hz}, 1 \mathrm{H}), 3.84-4.20(\mathrm{~m}, 5 \mathrm{H}), 4.59(\mathrm{dd}, J$ $=5.48 \mathrm{~Hz}, J=7.21 \mathrm{~Hz}, 1 \mathrm{H}), 4.69(\mathrm{~d}, J=11.97 \mathrm{~Hz}, 1 \mathrm{H}), 5.00$ (d, $J=11.97 \mathrm{~Hz}, 1 \mathrm{H}$ ), $5.75(\mathrm{~d}, J=5.99 \mathrm{~Hz}, 1 \mathrm{H}), 6.18$ (dd, $J$ $=15.9 \mathrm{~Hz}, J=5.72 \mathrm{~Hz}, 1 \mathrm{H}), 6.75(\mathrm{~d}, J=15.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.20-$ $7.44(\mathrm{~m}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 18.85,26.08,26.86,60.19$, $71.52,71.69,72.95,73.76,78.61,103.12,104.38$, 112.36 , $125.64,125.68,126.96,128.02,128.44,128.61,129.05,130.31$, $134.30,135.64,135.75,136.92$; IR $3059,1653,1604 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $438\left(\mathrm{M}^{+}, 2\right), 217$ (9), 171 (17), 131 (30), 115 (19), 105 (100), 91 (5), 69 (7). Anal. Calcd for $\mathrm{C}_{26} \mathrm{H}_{30} \mathrm{O}_{6}$ : C, 71.21; H, 6.89. Found: C, 70.97; H, 6.93.

1,2-O-Isopropylidene-3-O-( $m$-methylbenzyl)-4,5-O-[(1'R)-trans-3'-phenyl-2'-propen- $1^{\prime}$ 'yll $\boldsymbol{\beta}$-d-fructopyranose (31A): $56 \%$ as a white solid; [ $\alpha]_{\mathrm{D}}-123.8^{\circ}$ (c 0.084 , acetone); mp 87.1 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.43(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 3 \mathrm{H}), 2.25(\mathrm{~s}, 3$ $\mathrm{H}), 3.48(\mathrm{~d}, J=7.03 \mathrm{~Hz}, 1 \mathrm{H}), 3.86(\mathrm{~d}, J=8.38 \mathrm{~Hz}, 1 \mathrm{H}), 4.04-$ 4.20 (m, 4 H ), 4.43 (dd, $J=5.83 \mathrm{~Hz}, J=7.01 \mathrm{~Hz}, 1 \mathrm{H}), 4.65$ (d, $J=12.04 \mathrm{~Hz}, 1 \mathrm{H}), 4.90(\mathrm{~d}, J=12.04 \mathrm{~Hz}, 1 \mathrm{H}), 5.51(\mathrm{~d}, J$ $=6.64 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.11 (dd, $J=6.69 \mathrm{~Hz}, J=15.9 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.80 (d, $J=15.98 \mathrm{~Hz}, 1 \mathrm{H}$ ) , $7.06-7.43(\mathrm{~m}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) $21.32,26.01,26.94,59.82,71.82,72.76,75.89,76.04,77.42$, $104.40,112.50,125.07,125.63,127.04,128.20,128.38,128.61$, 128.67, 135.71, 138.10; IR $3030,1657 \mathrm{~cm}^{-1}$; MS (EI) $m / e$ (relative intensity) $438\left(\mathrm{M}^{+}, 7\right), 333(10), 260(21), 229$ (5), 194 (10), 171 (25), 131 (20), $100^{\circ}(100), 91$ (8), 77 (7). Anal. Calcd for $\mathrm{C}_{26} \mathrm{H}_{30} \mathrm{O}_{6}$ : $\mathrm{C}, 71.21 ; \mathrm{H}, 6.89$. Found: C, 71.47; H, 7.02 .

1,2-O-Isopropylidene-3-O-( $m$-methylbenzyl)-4,5-O-[(1'S)-trans-3'-phenyl- 2 'propen- $1^{\prime}$-yll- $\beta$-D-fructopyranose (31B): $37 \%$ as a white solid; $[\alpha]_{\mathrm{D}}-49.5^{\circ}\left(c 0.2\right.$, acetone); $\mathrm{mp} 82.4^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.44(\mathrm{~s}, 3 \mathrm{H}), 1.50(\mathrm{~s}, 3 \mathrm{H}), 2.35(\mathrm{~s}, 3 \mathrm{H})$, 3.52 (d, $J=7.47 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.89 (d, $J=8.40 \mathrm{~Hz}, 1 \mathrm{H}$ ), $3.99-$ 4.19 (m, 4 H ), 4.59 (dd, $J=5.25 \mathrm{~Hz}, J=7.48 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.69 (d, $J=11.82 \mathrm{~Hz}, 1 \mathrm{H}), 4.96(\mathrm{~d}, J=11.82 \mathrm{~Hz}, 1 \mathrm{H}), 5.67(\mathrm{~d}, J$ $=6.10 \mathrm{~Hz}, 1 \mathrm{H}), 6.15(\mathrm{dd}, J=6.10 \mathrm{~Hz}, J=15.85 \mathrm{~Hz}, 1 \mathrm{H})$, $6.73(\mathrm{~d}, J=15.85 \mathrm{~Hz}, 1 \mathrm{H}), 7.05-7.4(\mathrm{~m}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ${ }_{\left(\mathrm{CDCl}_{3}\right)}$ 21.40, 26.02, 26.93, 60.17, 71.78, 72.54, 73.07, 73.78, $78.56,103.15,104.39,112.40,125.13,125.63,126.95,128.29$,
$128.44,128.49,128.60,128.86,134.36,135.65,137.77$; IR 2987, $1658 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $438\left(\mathrm{M}^{+}, 13\right), 423$ (11). 217 (4), 194 (4), 171 (13), 145 (6), 131 (24), 115 (15), 105 (100), 77 (6), 69 (8). Anal. Calcd for $\mathrm{C}_{26} \mathrm{H}_{30} \mathrm{O}_{6}: \mathrm{C}, 71.21 ; \mathrm{H}$, 6.89. Found: C, 70.97; H, 6.93.

1,2-O-Isopropylidene-3-O-(p-methylbenzyl)-4,5-O-[(1'R)-trans-3'-phenyl- $2^{\prime}$-propen- $\left.1^{\prime}-\mathrm{yl}\right]-\beta$-d-fructopyranose (32A): $51 \%$ as a white solid; $[\alpha]_{D}-132^{\circ}$ (c 0.05 , acetone); mp 122.1 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.41(\mathrm{~s}, 3 \mathrm{H}), 1.47(\mathrm{~s}, 3 \mathrm{H}), 2.31(\mathrm{~s}, 3$ $\mathrm{H}), 3.47(\mathrm{~d}, J=7.14 \mathrm{~Hz}, 1 \mathrm{H}), 3.84(\mathrm{~d}, J=8.46 \mathrm{~Hz}, 1 \mathrm{H}), 4.03$ $-4.19(\mathrm{~m}, 4 \mathrm{H}), 4.42(\mathrm{dd}, J=5.75 \mathrm{~Hz}, J=7.05 \mathrm{~Hz}, 1 \mathrm{H}), 4.65$ $(\mathrm{d}, J=11.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.89(\mathrm{~d}, J=11.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.51(\mathrm{~d}, J=$ $6.58 \mathrm{~Hz}, 1 \mathrm{H}), 6.10(\mathrm{dd}, J=6.61 \mathrm{~Hz}, J=15.94 \mathrm{~Hz}, 1 \mathrm{H}), 6.78$ (d, $J=15.94 \mathrm{~Hz}, 1 \mathrm{H}), 7.03-7.43(\mathrm{~m}, 9 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR: 21.13, $26.02,26.94,59.82,71.82,72.50,75.49,76.06,77.44,104.41$, $112.28,125.79,127.05,128.22,128.61,128.98,134.83,135.61$, 137.35 ; IR $2994 \mathrm{~cm}^{-1}$; MS (EI) $m / e$ (relative intensity) 438 ( $\mathrm{M}^{+}$, 1), 260 (4), 194 (19), 171 (29), 131 (22), 105 (100), 91 (5), 77 (4). Anal. Calcd for $\mathrm{C}_{26} \mathrm{H}_{30} \mathrm{O}_{6}$ : C, 71.21; H, 6.89. Found: C, 71.22; H, 6.94.

1,2-O-Isopropylidene-3-O-( $p$-methylbenzyl)-4,5-O-[(1'S)-trans-3'-phenyl- $2^{\prime}$-propen- $1^{\prime}-$ yl] $-\beta$-D-fructopyranose (32B): $36 \%$ as a white solid; $[\alpha]_{D}-54.7^{\circ}$ (c 0.137 , acetone); mp $75{ }^{\circ} \mathrm{C}$; ${ }^{1}{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.42(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 3 \mathrm{H}), 2.33(\mathrm{~s}, 3 \mathrm{H})$, $3.50(\mathrm{~d}, J=7.44 \mathrm{~Hz}, 1 \mathrm{H}), 3.86$ (d, $J=8.54 \mathrm{~Hz}, 1 \mathrm{H}), 3.97-$ $4.18(\mathrm{~m}, 4 \mathrm{H}), 4.57$ (dd, $J=5.27 \mathrm{~Hz}, J=7.47 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.69 (d, $J=11.73 \mathrm{~Hz}, 1 \mathrm{H}$ ), $4.92(\mathrm{~d}, J=11.73 \mathrm{~Hz}, 1 \mathrm{H}), 5.66(\mathrm{~d}, J$ $=6.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.14(\mathrm{dd}, J=6.19 \mathrm{~Hz}, J=15.89 \mathrm{~Hz}, 1 \mathrm{H}), 6.72$ (d, $J=15.90 \mathrm{~Hz}, 1 \mathrm{H}), 7.12-7.43(\mathrm{~m}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) $\delta 21.16,26.01,26.92,60.14,71.73,72.33,72.92,73.76,78.57$, $103.10,104.38,112.36,125.62,126.95,128.22,128.42,128.59$, 129.04, 134.32, 134.76, 135.64; IR 2987, $1657 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $438\left(\mathrm{M}^{+}, 1\right), 277$ (1), 217 (6), 171 (14), 131 (20), 115 (12), 105 (100), 69 (14), 43 (25). Anal. Calcd for $\mathrm{C}_{26} \mathrm{H}_{30} \mathrm{O}_{6}$ : C, 71.21; H, 6.89. Found: C, 71.29; H, 6.91.

1,2-O-Isopropylidene-3-O-(p-methoxybenzyl)-4,5-O[( $1^{\prime} R$ )-trans-3'-phenyl-2'-propen-1'-yl]- $\beta$-D-fructopyranose (33A): $54 \%$ as a white solid; $[\alpha]_{D}-119.2^{\circ}$ (c 0.052 , acetone); $\mathrm{mp} 93.5^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ) $\delta 1.41$ (s, 3 H ), 1.48 (s, 3 H ), 3.46 (d, $J=7.18 \mathrm{~Hz}, 1 \mathrm{H}$ ), $3.74(\mathrm{~s}, 3 \mathrm{H}), 3.79-4.16$ (m, $5 \mathrm{H}), 4.43$ (dd, $J=5.79 \mathrm{~Hz}, J=7.17 \mathrm{~Hz}, 1 \mathrm{H}), 4.63(\mathrm{~d}, J=$ $11.78 \mathrm{~Hz}, 1 \mathrm{H}), 4.85(\mathrm{~d}, J=11.78 \mathrm{~Hz}, 1 \mathrm{H}), 5.52(\mathrm{~d}, J=6.74$ $\mathrm{Hz}, 1 \mathrm{H}$ ), 6.13 (dd, $J=6.70 \mathrm{~Hz}, J=15.98 \mathrm{~Hz}, 1 \mathrm{H}), 6.80(\mathrm{~d}, J$ $=15.98 \mathrm{~Hz}, 1 \mathrm{H}), 6.78(\mathrm{~d}, J=8.68 \mathrm{~Hz}, 2 \mathrm{H}), 7.22-7.43(\mathrm{~m}, 7$ $\mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 26.03,26.92,55.21,59.93,71.83,72.27$, $75.29,76.08,77.49,104.38,104.43,112.28,113.71,125.81$, $126.97,127.04,128.59,128.64,129.74,129.96$, 135.59; IR 3034, $1612 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $454\left(\mathrm{M}^{+}, 1\right), 260$ (10), 171 (18), 121 (100), 115 (9), 91 (3), 77 (3). Anal. Calcd for $\mathrm{C}_{26} \mathrm{H}_{30} \mathrm{O}_{7}$ : C, 68.71; H, 6.65. Found: C, $68.58 ; \mathrm{H}, 6.81$.

1,2-O-Isopropylidene-3-O-(p-methoxybenzyl)-4,5-O. [( $1^{\prime} S$ )-trans-3'-phenyl-2'-propen-1'-yl]- $\beta$-d-fructopyranose (33B): $34 \%$ as a white solid; $[\alpha]_{D}-50^{\circ}$ (c 0.112 , acetone); $\mathrm{mp} 82.5{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.41(\mathrm{~s}, 3 \mathrm{H}), 1.48(\mathrm{~s}, 3 \mathrm{H})$, $3.49(\mathrm{~d}, \mathrm{~J}=7.43 \mathrm{~Hz}, 1 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 3.77-4.18(\mathrm{~m}, 5 \mathrm{H})$, 4.56 (dd, $J=5.20 \mathrm{~Hz}, J=7.50 \mathrm{~Hz}, 1 \mathrm{H}), 4.66(\mathrm{~d}, J=11.62$ $\mathrm{Hz}, 1 \mathrm{H}), 4.88(\mathrm{~d}, J=11.62 \mathrm{~Hz}, 1 \mathrm{H}), 5.64(\mathrm{~d}, J=6.09 \mathrm{~Hz}, 1$ $\mathrm{H}), 6.14(\mathrm{dd}, J=6.15 \mathrm{~Hz}, J=15.93 \mathrm{~Hz}, 1 \mathrm{H}), 6.72(\mathrm{~d}, J=$ $15.93 \mathrm{~Hz}, 1 \mathrm{H}), 6.86(\mathrm{~d}, J=8.69 \mathrm{~Hz}, 2 \mathrm{H}), 7.24-7.43(\mathrm{~m}, 7 \mathrm{H})$; ${ }^{13}{ }^{1} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 25.99,26.89,55.25,60.13,71.71,72.09$, $72.73,73.76,78.55,103.09,104.39,112.35,113.77,125.59$,

Table 9. Simmons-Smith Reactions of Psicopyranose Derivatives

| starting material | cyclopropanation |  |  | $\frac{\text { hydrol-redn }}{\text { yield (\%) }}$ | product |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | temp ( ${ }^{\circ} \mathrm{C}$ ) | time (h) | yield (\%) |  | $\mathrm{ee}(\%)^{a}$ | config ${ }^{\text {b }}$ |
| endo isomer 41A | $-30 \rightarrow 0$ | 48 | 82 | 90 | 48 | (1R,2R) |
| exo isomer 41B | $-30 \rightarrow 0$ | 48 | 64 | 82 | 20 | $(1 S, 2 S)$ |

${ }^{a}$ HPLC analysis (Daicel OJ, 10\% i-PrOH in hexane). ${ }^{b} 49:[\alpha]_{\mathrm{D}}-41.7^{\circ}(c 0.42, \mathrm{EtOH})\left\{\mathrm{lit} .{ }^{11 \mathrm{c}}[\alpha]_{\mathrm{D}}-92^{\circ}(c \mathrm{l} .23, \mathrm{EtOH})\right\} .64:[\alpha]_{\mathrm{D}}+19^{\circ}$ (c $0.1, \mathrm{EtOH}$ ).

126.94, 128.43, 128.59, 129.74, 129.85, 134.35, 135.62, 159.31; IR $3034,1613 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) 454 (M ${ }^{+}$, 1), 318 (1), 260 (2), 217 ( 8 ), 171 (15), 148 (10), 131 (22), 121 (120), 104 (5), 69 (5). Anal. Calcd for $\mathrm{C}_{26} \mathrm{H}_{30} \mathrm{O}_{7}$ : C, $68.71 ; \mathrm{H}$, 6.65. Found: C, 68.31; H, 6.79.

1,2-Isopropylidene-3-O-(p-tert-butylbenzyl)-4,5-O-[( $\mathbf{1}^{\prime} R$ ). trans-3'-phenyl-2'-propen- $1^{\prime}$-yll- $\boldsymbol{\beta}$-d-fructopyranose (34A): $49 \%$ as a white solid; $[\alpha]_{\mathrm{D}}-112.4^{\circ}$ (c 0.121 , acetone); mp 116.3 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ) $\delta 1.29(\mathrm{~s}, 9 \mathrm{H}), 1.44(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 3$ H), $3.48(\mathrm{~d}, J=7.13 \mathrm{~Hz}, 1 \mathrm{H}), 3.84(\mathrm{~d}, J=8.41 \mathrm{~Hz}, 1 \mathrm{H}), 4.05-$ $4.20(\mathrm{~m}, 4 \mathrm{H}), 4.45(\mathrm{dd}, J=6.67 \mathrm{~Hz}, J=7.11 \mathrm{~Hz}, 1 \mathrm{H}), 4.68$ (d, $J=11.99 \mathrm{~Hz}, 1 \mathrm{H}), 4.89(\mathrm{~d}, J=11.99 \mathrm{~Hz}, 1 \mathrm{H}), 5.52(\mathrm{~d}, J$ $=6.61 \mathrm{~Hz}, 1 \mathrm{H}), 6.18(\mathrm{dd}, J=6.60 \mathrm{~Hz}, J=15.93 \mathrm{~Hz}, 1 \mathrm{H})$, $6.80(\mathrm{~d}, J=15.95 \mathrm{~Hz}, 1 \mathrm{H}), 7.24-7.44(\mathrm{~m}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 26.08,26.92,31.34,59.82,71.82,72.33,75.42,76.08$, $77.52,104.45,112.28,125.19,125.73,127.05,127.98,128.65$, 134.83, 135.70; IR 2962, $1653 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) 480 ( $\mathrm{M}^{+}, 17$ ), 318 (15), 236 (11), 171 (31), 147 (100), 131 (26), 115 (22), 91 (14), 69 (8), 59 (15). Anal. Calcd for $\mathrm{C}_{29} \mathrm{H}_{36} \mathrm{O}_{6}: ~ \mathrm{C}, 72.48 ; \mathrm{H}, 7.55$. Found: C, 72.44; $\mathrm{H}, 7.56$.

1,2-Isopropylidene-3-O-(p-tert-butylbenzyl)-4,5-O-[(1'S)-trans-3'-phenyl-2'-propen- $1^{\prime}$-yll- $\beta$-D-fructopyranose (34B): $41 \%$ as a white solid; $[\alpha]_{D}-54.4^{\circ}$ (c 0.103 , acetone); mp 75.6 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.30(\mathrm{~s}, 9 \mathrm{H}), 1.45(\mathrm{~s}, 3 \mathrm{H}), 1.51(\mathrm{~s}, 3$ H), $3.53(\mathrm{~d}, J=7.61 \mathrm{~Hz}, 1 \mathrm{H}), 3.88(\mathrm{~d}, J=8.92 \mathrm{~Hz}, 1 \mathrm{H}), 4.01-$ $4.21(\mathrm{~m}, 4 \mathrm{H}), 4.60(\mathrm{dd}, J=6.23 \mathrm{~Hz}, J=7.60 \mathrm{~Hz}, 1 \mathrm{H}), 4.73$ (d, $J=11.85 \mathrm{~Hz}, 1 \mathrm{H}), 4.95(\mathrm{~d}, J=11.85 \mathrm{~Hz}, 1 \mathrm{H}), 5.68(\mathrm{~d}, J$ $=6.06 \mathrm{~Hz}, 1 \mathrm{H}), 6.20(\mathrm{dd}, J=6.03 \mathrm{~Hz}, J=15.89 \mathrm{~Hz}, 1 \mathrm{H})$, $6.74(\mathrm{~d}, J=15.90 \mathrm{~Hz}, 1 \mathrm{H}), 7.28-7.45(\mathrm{~m}, 9 \mathrm{~Hz})$; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 26.06,26.89,31.35,60.15,71.75,72.37,72.81,73.76$, $78.57,103.09,104.40,112.37,125.28,125.60,126.95,127.86$, 128.31, 128.42, 128.59, 128.93, 134.32, 134.76; IR 3030, 1676, $1628 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $480\left(\mathrm{M}^{+}, 2\right), 465$ (1), 333 (1), 260 (1), 217 (12), 171 (23), 147 (100), 131 (35), 115 (20), 91 (9), 69 (12). Anal. Calcd for $\mathrm{C}_{29} \mathrm{H}_{36} \mathrm{O}_{6}$ : $\mathrm{C}, 72.48 ; \mathrm{H}$, 7.55. Found: C, 72.66; H, 7.41.

1,2-O-Isopropylidene-3-O-(p-phenylbenzyl)-4,5-O-[(1'R). trans-3'-phenyl-2'-propen- $1^{\prime}$ - yl 1$]-\beta$-D-fructopyranose (35A): $55 \%$ as a white solid; $[\alpha]_{\mathrm{D}}-128.6^{\circ}$ (c 0.091 , acetone); mp 121.8 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.45(\mathrm{~s}, 3 \mathrm{H}), 1.50(\mathrm{~s}, 3 \mathrm{H}), 3.52(\mathrm{~d}, J=$ $7.22 \mathrm{~Hz}, 1 \mathrm{H}), 3.90(\mathrm{~d}, J=8.47 \mathrm{~Hz}, 1 \mathrm{H}), 4.05-4.22(\mathrm{~m}, 4 \mathrm{H})$, 4.46 (dd, $J=5.86 \mathrm{~Hz}, J=6.97 \mathrm{~Hz}, 1 \mathrm{H}), 4.75$ (d, $J=12.14$ $\mathrm{Hz}, 1 \mathrm{H}), 4.98(\mathrm{~d}, J=12.14 \mathrm{~Hz}, 1 \mathrm{H}), 5.53(\mathrm{~d}, J=6.60 \mathrm{~Hz}, 1$ H), 6.12 (dd, $J=6.66 \mathrm{~Hz}, J=15.95 \mathrm{~Hz}, 1 \mathrm{H}), 6.79(\mathrm{~d}, J=$ $15.94 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.24-7.57(\mathrm{~m}, 14 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) 26.72 , $27.55,60.48,72.52,72.90,76.72,78.03,78.27$. 105.08, 112.99, $126.35,127.66,129.18,129.25,129.38,136.31,137.55$; IR 3030, $1655 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $500\left(\mathrm{M}^{+}, 18\right), 318$ (11), 300 (20), 260 (8), 167 (100), 131 (14), 115 (14), 104 (5), 69 (5). Anal. Calcd for $\mathrm{C}_{31} \mathrm{H}_{32} \mathrm{O}_{6}$ : C, 74.38; H, 6.44. Found: C, 74.34; H, 6.49.

1,2-O-Isopropylidene-3-O-(p-phenylbenzyl)-4,5-O-[(1'S)-trans-3'-phenyl- $\mathbf{2}^{\prime}$-propen- $1^{\prime}$-yll- $\boldsymbol{\beta}$-d-fructopyranose (35B): $35 \%$ as a white solid; $[\alpha]_{\mathrm{D}}-56.9^{\circ}$ (c 0.065 , acetone); mp 100.8 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.45(\mathrm{~s}, 3 \mathrm{H}), 1.50(\mathrm{~s}, 3 \mathrm{H}), 3.56(\mathrm{~d}, \mathrm{~J}=$ $7.63 \mathrm{~Hz}, 1 \mathrm{H}), 3.93(\mathrm{~d}, J=8.42 \mathrm{~Hz}, 1 \mathrm{H}), 4.06-4.20(\mathrm{~m}, 4 \mathrm{H})$, 4.59 (dd, $J=5.11 \mathrm{~Hz}, J=7.53 \mathrm{~Hz}, 1 \mathrm{H}), 4.78(\mathrm{~d}, J=11.99$ $\mathrm{Hz}, 1 \mathrm{H}), 5.01(\mathrm{~d}, J=11.99 \mathrm{~Hz}, 1 \mathrm{H}), 5.68(\mathrm{~d}, J=6.05 \mathrm{~Hz}, 1$ $\mathrm{H}), 6.15$ (dd, $J=6.25 \mathrm{~Hz}, J=15.87 \mathrm{~Hz}, 1 \mathrm{H}), 6.71(\mathrm{~d}, J=$ $15.87 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.24-7.61(\mathrm{~m}, 14 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) 26.06 , $26.89,60.17,71.82,72.81,73.76,78.47,103.15,104.39,112.43$, $125.55,126.95,127.09,127.30,128.49,128.58,128.77,134.37$, 135.59, 136.87; IR $3030,1653 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $500\left(\mathrm{M}^{+}, 1\right), 260(2), 217(6), 167$ (100), 152 ( 8 ), 131 (4), 115 (14), 104 (5), 69 (8), 43 (5). Anal. Calcd for $\mathrm{C}_{31} \mathrm{H}_{32} \mathrm{O}_{6}$ : C, 74.38 ; H, 6.44. Found: C, 74.33 ; H, 6.51 .

1,2-O-Isopropylidene-3-O-(2-naphthylmethyl)-4,5-O. [(1'R)-trans-3'-phenyl-2'-propen-1'-yl]- $\beta$-D-fructopyranose (36A): $51 \%$ as a white solid; $[\alpha]_{\mathrm{D}}-141.2^{\circ}$ (c 0.102 , acetone); mp $146.8{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ) $\delta 1.46$ (s, 3 H ), 1.50 (s, 3 H ), $3.52(\mathrm{~d}, J=7.14 \mathrm{~Hz}, 1 \mathrm{H}), 3.84(\mathrm{~d}, J=8.47 \mathrm{~Hz}, 1 \mathrm{H})$, $4.04-4.21(\mathrm{~m}, 4 \mathrm{H}), 4.49(\mathrm{dd}, J=6.63 \mathrm{~Hz}, J=7.04 \mathrm{~Hz}, 1 \mathrm{H})$, $4.88(\mathrm{~d}, J=12.14 \mathrm{~Hz}, 1 \mathrm{H}), 5.08(\mathrm{~d}, J=12.14 \mathrm{~Hz}, 1 \mathrm{H}), 5.54$ (d, $J=6.54 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.11 (dd, $J=6.57 \mathrm{~Hz}, J=15.96 \mathrm{~Hz}, 1$ $\mathrm{H}), 6.81(\mathrm{~d}, J=15.96 \mathrm{~Hz}, 1 \mathrm{H}), 7.30-7.79(\mathrm{~m}, 12 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 26.07,26.93,59.81,71.89,72.59,75.60,76.06,77.50$, $104.36,104.47,112.36,125.63,125.91,126.06,127.11,127.63$, 127.96, 128.13, $128.59,128.66,132.96,133.11,135.26,135.55$, 135.81; IR $3055,1662 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $474\left(\mathrm{M}^{+}, 2\right), 260(8), 171(22), 141$ (100), 115 (26), 104 (6), 91 (5), 69 (6), 43 (10). Anal. Calcd for $\mathrm{C}_{29} \mathrm{H}_{30} \mathrm{O}_{6}$ : C, $73.40 ; \mathrm{H}$, 6.37. Found: C, 73.38; H, 6.43 .

1,2-O-Isopropylidene-3-O-(2-naphthylmethyl)-4,5-O-[(1'S)-trans-3'-phenyl-2'-propen-1'-yl]- $\beta$-D-fructopyranose (36B): $37 \%$ as a white solid; $[\alpha]_{\mathrm{D}}-60^{\circ}(c) 0.1$, acetone); $\mathrm{mp} 87.3{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.46(\mathrm{~s}, 3 \mathrm{H}), 1.50(\mathrm{~s}, 3 \mathrm{H})$, $3.55(\mathrm{~d}, J=7.54 \mathrm{~Hz}, 1 \mathrm{H}), 3.89(\mathrm{~d}, J=8.41 \mathrm{~Hz}, 1 \mathrm{H}), 4.0-4.19$ ( $\mathrm{m}, 4 \mathrm{H}$ ), 4.61 (dd, $J=5.29 \mathrm{~Hz}, J=7.45 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.90 (d, $J$ $=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.11(\mathrm{~d}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.59(\mathrm{~d}, J=6.19$ $\mathrm{Hz}, 1 \mathrm{H}), 6.14$ (dd, $J=6.24 \mathrm{~Hz}, J=15.85 \mathrm{~Hz}, 1 \mathrm{H}), 6.63$ (d, $J$ $=15.84 \mathrm{~Hz}, 1 \mathrm{H}), 7.24-7.79(\mathrm{~m}, 12 \mathrm{H}) ;{ }^{13} \mathrm{C}^{\mathrm{NMR}}\left(\mathrm{CDCl}_{3}\right) 26.06$, $26.91,60.17,71.80,72.40,73.13,73.78,78.50,103.19,104.38$, $112.43,125.46,126.04,126.17,126.97$, 127.13, 127.72, 127.88 , $128.21,128.45,128.60,133.03,133.16,134.48,135.20,135.60$; IR $3053,1657 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) 474 ( $\mathrm{M}^{+}$, 2), 217 (4), 141 (100), 131 (15), 115 (25), 91 (5), 69 (18), 43 (25). Anal. Caled for $\mathrm{C}_{29} \mathrm{H}_{30} \mathrm{O}_{6}$ : C, 73.40; H, 6.37. Found: C, 73.41; H, 6.49.

1,2-O-Isopropylidene-3-O-(p-phenylbenzyl)-4,5-O-[(1'R)-trans-5'-phenyl- $2^{\prime}$-penten- $1^{\prime}$-yll- $\beta$-d-fructopyranose ( 37 A ): $53 \%$ as a white solid; [ $\alpha]_{\mathrm{D}}-54.2^{\circ}$ (c 0.059 , acetone); mp 104.5 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.43(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 3 \mathrm{H}), 2.42(\mathrm{~m}, 2$ H ), 2.74 (t, 2 H ), $3.45(\mathrm{~d}, J=7.12 \mathrm{~Hz}, 1 \mathrm{H}), 3.88(\mathrm{~d}, J=8.5$ $\mathrm{Hz}, 1 \mathrm{H}), 4.08-4.14(\mathrm{~m}, 4 \mathrm{H}), 4.40(\mathrm{dd}, J=6.91 \mathrm{~Hz}, J=7.08$ $\mathrm{Hz}, 1 \mathrm{H}), 4.67(\mathrm{~d}, J=12.05 \mathrm{~Hz}, 1 \mathrm{H}), 4.95(\mathrm{~d}, J=12.05 \mathrm{~Hz}, 1$ H), 5.32 (d, $J=6.90 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.52 (dd, $J=6.88 \mathrm{~Hz}, J=15.9$ $\mathrm{Hz}, 1 \mathrm{H}), 6.02(\mathrm{dt}, J=15.95 \mathrm{~Hz}, 1 \mathrm{H}), 7.14-7.59(\mathrm{~m}, 14 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) 26.07, 26.91, 33.77, 34.95, 59.87, 71.69, $72.50,75.84,76.31,77.22,104.37,104.51,112.30,125.99$, $127.02,127.06,127.26,127.58,128.36,128.75,137.13,137.48$, $140.58,140.82,141.28$; IR $3026,1674 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $528\left(\mathrm{M}^{+}\right), 288$ (5), 199 (10), 181 (4), 167 (100), 117 (16), 91 (75), 69 (10), 43 (18). Anal. Caled for $\mathrm{C}_{33} \mathrm{H}_{36} \mathrm{O}_{6}$ : C, $75.0 ; \mathrm{H}, 6.86$. Found: C, 75.09 ; H, 7.04 .

1,2-O-Isopropylidene-3-O-( $p$-phenylbenzyl)-4,5-O-[(1'S)-trans-5'.phenyl- $2^{\prime}$-penten- $\left.1^{\prime} \cdot \mathrm{yl}\right]-\beta$-D-fructopyranose (37B): $35 \%$ as an oily solid; [ $\alpha$ ] $-38.5^{\circ}$ (c 0.039 , acetone); ${ }^{1} \mathrm{H}$ NMR
$\left(\mathrm{CDCl}_{3}\right) \delta 1.44(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 3 \mathrm{H}), 2.41(\mathrm{~m}, 2 \mathrm{H}), 2.71(\mathrm{t}, 2$ $\mathrm{H}), 3.50(\mathrm{~d}, J=7.54 \mathrm{~Hz}, 1 \mathrm{H}), 3.89(\mathrm{~d}, J=8.59 \mathrm{~Hz}, 1 \mathrm{H}), 4.01-$ 4.16 (m, 4 H ), 4.54 (dd, $J=5.16 \mathrm{~Hz}, J=7.53 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.73 (d, $J=11.99 \mathrm{~Hz}, 1 \mathrm{H}), 4.98(\mathrm{~d}, J=11.99 \mathrm{~Hz}, 1 \mathrm{H}), 5.45(\mathrm{~d}, J$ $=6.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.53(\mathrm{dd}, J=14.46 \mathrm{~Hz}, 1 \mathrm{H}), 5.94(\mathrm{dt}, J=$ $14.47 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.14-7.60(\mathrm{~m}, 14 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) 26.08 , $26.90,33.73,35.03,60.23,71.83,72.53,72.73,73.68,78.43$, $103.34,104.36,112.41,125.98,127.09,127.31,127.38,128.37$, $128.49,128.76,136.25,136.89$; IR $3059,1672 \mathrm{~cm}^{-1}$; MS (EI) $m / e$ (relative intensity) $528\left(\mathrm{M}^{+}\right), 320(1), 245$ (4), 167 (100), 115 (24), 91 (54), 69 (11), 43 (14). Anal. Calcd for $\mathrm{C}_{33} \mathrm{H}_{36} \mathrm{O}_{6}$ : C, 75.0; H, 6.86. Found: C, 74.77; H, 6.93 .

1,2-O-Isopropylidene-3-O-( $p$-phenylbenzyl)-4,5-O-[( $\left.1^{\prime} R\right)$ -trans-4'-(benzyloxy)-2'-buten- $1^{\prime}$-yll- $\beta$-D-fructopyranose (38A): $47 \%$ as a white solid; $[\alpha]_{D}-94.5^{\circ}$ (c 0.055, acetone); $\mathrm{mp} 75.2{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.43(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 3 \mathrm{H})$, 3.47 (d, $J=7.17 \mathrm{~Hz}, 1 \mathrm{H}), 3.87(\mathrm{~d}, J=8.53 \mathrm{~Hz}, 1 \mathrm{H}), 4.09(\mathrm{~m}$, $6 \mathrm{H}), 4.42$ (dd, $J=5.74 \mathrm{~Hz}, J=7.08 \mathrm{~Hz}, 1 \mathrm{H}), 4.45(\mathrm{~s}, 2 \mathrm{H})$, $4.69(\mathrm{~d}, J=11.99 \mathrm{~Hz}, 1 \mathrm{H}), 4.97(\mathrm{~d}, J=11.99 \mathrm{~Hz}, 1 \mathrm{H}), 5.39$ (d, $J=6.42 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.84 (ddt, $J=6.40 \mathrm{~Hz}, J=15.58 \mathrm{~Hz}, 1$ $\mathrm{H}), 6.09$ (dt, $J=15.59 \mathrm{~Hz}, 1 \mathrm{H}), 7.23-7.58(\mathrm{~m}, 14 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 26.07,26.90,59.86,69.29,71.87,72.48,72.63,75.97$, $76.37,77.32,103.74,104.36,112.32,127.04,127.26,127.65$, 127.68, 128.32, 128.42, 128.66, 128.74, 133.71; IR 2934, 1601 $\mathrm{cm}^{-1}$; MS (EI) m/e (relative intensity) 544 ( $\mathrm{M}^{+}$), 436 (1), 254 (2), 167 (74), 91 (100), 69 (10). Anal. Calcd for $\mathrm{C}_{33} \mathrm{H}_{36} \mathrm{O}_{7}$ : C, 72.77; H, 6.66. Found: C, 72.42; H, 6.75 .

1,2-O-Isopropylidene-3-O- $(p$-phenylbenzyl)-4,5-O-[(1'S)-trans-4'-(benzyloxy)-2'-buten-1'-yl]- $\beta$-d-fructopyranose (38B): $43 \%$ as an oily solid; $[\alpha]_{D}-68.8^{\circ}\left(c 0.236\right.$, acetone); ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.45(\mathrm{~s}, 3 \mathrm{H}), 1.50(\mathrm{~s}, 3 \mathrm{H}), 3.52(\mathrm{~d}, J=7.48$ $\mathrm{Hz}, 1 \mathrm{H}$ ), $3.90(\mathrm{~d}, J=8.52 \mathrm{~Hz}, 1 \mathrm{H}), 4.08(\mathrm{~m}, 6 \mathrm{H}), 4.55(\mathrm{~s}, 2$ $\mathrm{H}), 4.56(\mathrm{~m}, 1 \mathrm{H}), 4.75$ (d, $J=11.96 \mathrm{~Hz}, 1 \mathrm{H}), 5.00(\mathrm{~d}, J=$ $11.96 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.58 (d, $J=5.92 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.77 (dd, $J=5.98$ $\mathrm{Hz}, J=15 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.02 (dt, $J=15 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.30-7.61$ (m, $14 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) $26.06,26.89,60.14,69.28,71.82$, $72.52,72.76,72.84,73.68,78.43,102.53,104.36,112.41$, $127.08,127.29,127.68,128.41,128.75,132.16$; IR 2930,1689 $\mathrm{cm}^{-1}$; MS (EI) m/e (relative intensity) 544 ( $\mathrm{M}^{+}$), 436 (2), 362 (0.2), 309 (0.2), 273 (0.4), 167 (100), 91 (89), 69 (29). Anal. Calcd for $\mathrm{C}_{33} \mathrm{H}_{36} \mathrm{O}_{7}$ : C, 72.77; H, 6.66. Found; C, $73.08 ; \mathrm{H}$, 6.95.

1,2-O-Isopropylidene-3-O-(p-phenylbenzyl)-4,5-O-[(1'R)-$3^{\prime}$-methyl-2'-buten-1'-yl]- $\beta$-d-fructopyranose (39A): $50 \%$ as a white solid; $[\alpha]_{D}-82.2^{\circ}$ (c 0.107 , acetone); mp $104.2^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ) $\delta 1.43(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 3 \mathrm{H}), 1.78(\mathrm{~s}, 6 \mathrm{H}), 3.47$ (d, $J=7.06 \mathrm{~Hz}, \mathrm{H}_{3}$ ), $3.87(\mathrm{~d}, J=8.47 \mathrm{~Hz}, 1 \mathrm{H}), 4.01-4.13(\mathrm{~m}$, $4 \mathrm{H}), 4.39(\mathrm{t}, J=6.78 \mathrm{~Hz}, 1 \mathrm{H}), 4.70(\mathrm{~d}, J=12.05 \mathrm{~Hz}, 1 \mathrm{H})$, 4.96 (d, $J=12.05 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.27 (dm, 1 H ), 5.67 (d, $J=7.6$ $\mathrm{Hz}, 1 \mathrm{H}), 7.32-7.60(\mathrm{~m}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) 18.37, 25.95, $26.08,26.91,59.92,71.91,72.53,75.74,76.43,78.43,100.63$, $104.38,112.25,122.23,127.01,127.07,127.24,128.38,128.74$, 137.20, 140.89, 141.68; IR $2991 \mathrm{~cm}^{-1}$; MS (EI) $m / e$ (relative intensity) $452\left(\mathrm{M}^{+}, 10\right) 252(2), 212(11), 167(100), 123(22)$, 85 (11), 69 (10). Anal. Calcd for $\mathrm{C}_{27} \mathrm{H}_{32} \mathrm{O}_{6}$ : $\mathrm{C}, 71.66 ; \mathrm{H}, 7.13$. Found: C, 71.69; H, 6.59.

1,2-O-Isopropylidene-3- $O$-(p-phenylbenzyl)-4,5-O-[(1'R)-$3^{\prime}$-methyl-2'-buten- $1^{\prime}$-yl]- $\beta$-d-fructopyranose (39B): $28 \%$ as a white solid; $[\alpha]_{D}-99.7^{\circ}$ (c 0.305 , acetone); $\mathrm{mp} 54.2^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ) $\delta 1.44(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 3 \mathrm{H}), 1.78(\mathrm{~s}, 6 \mathrm{H}), 3.53$ (d, $J=7.63 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.90 (d, $J=8.37 \mathrm{~Hz}, 1 \mathrm{H}$ ), $3.95-4.17$ (m, 4 H ), 4.56 (dd, $J=5.19 \mathrm{~Hz}, J=7.49 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.76 (d, $J$ $=12.02 \mathrm{~Hz}, 1 \mathrm{H}), 4.99(\mathrm{~d}, J=12.02 \mathrm{~Hz}, 1 \mathrm{H}), 5.25(\mathrm{dm}, J=$ $7.82 \mathrm{~Hz}, 1 \mathrm{H}), 5.84(\mathrm{~d}, J=7.85 \mathrm{~Hz}, 1 \mathrm{H}), 7.33-7.60(\mathrm{~m}, 9 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) 18.24, 25.97, 26.09, 26.90, 60.37, 71.87, $72.14,72.57,73.71,78.45,99.51,104.38,112.20,122.24$, $127.10,127.13,127.32,128.53,128.77,136.91$; IR $2978 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $452\left(\mathrm{M}^{+}, 5\right), 225$ (3), 217 (12), 212 (12), 167 (100), 123 (17), 105 (20), 83 (9), 69 (5). Anal. Calcd for $\mathrm{C}_{27} \mathrm{H}_{32} \mathrm{O}_{6}$ : $\mathrm{C}, 71.66 ; \mathrm{H}, 7.13$. Found: C, $71.85 ; \mathrm{H}$, 7.37.

1,2-O-Isopropylidene-3-O-(p-phenylbenzyl)-4,5-O-[( $1^{\prime} R$ )-trans-2'-octen-1'-yll- $\beta$-d-fructopyranose (40A): $51 \%$ as a white solid; $[\alpha]_{\mathrm{D}}-87.4^{\circ}(c 0.154$, acetone $) ; \mathrm{mp} 74.1^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 0.86(\mathrm{t}, 3 \mathrm{H}), 1.25-1.55(\mathrm{~m}, 6 \mathrm{H}), 1.43(\mathrm{~s}, 3 \mathrm{H}), 1.50$ (s 3 H ), 2.09 (m, 2 H ), 3.47 (d, $J=7.09 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.88 (d, $J=$
$8.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.09(\mathrm{~m}, 4 \mathrm{H}), 4.39(\mathrm{t}, J=7.08 \mathrm{~Hz}, 1 \mathrm{H}), 4.70(\mathrm{~d}$, $J=11.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.97(\mathrm{~d}, J=11.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.32(\mathrm{~d}, J=7.05$ $\mathrm{Hz}, 1 \mathrm{H}$ ), 5.48 (ddt, $J=7.01 \mathrm{~Hz}, J=15 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.98 (dt, $J$ $=15.24 \mathrm{~Hz}, 1 \mathrm{H}), 7.32-7.60(\mathrm{~m}, 9 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta$ 14.01, 22.49, 26.09, 26.91, 28.37, 31.40, 32.03, 59.91, 71.92, $72.54,75.85,77.19,104.40,104.82,112.30,126.86,127.04$, 127.08, 127.28, 128.40, 128.76, 137.17, 138.87, 140.61; IR 2987 $\mathrm{cm}^{-1}$; MS (EI) m/e (relative intensity) 494 (M+), 254 (13), 211 (5), 167 (100), 109 (3), 95 (12), 69 (15), 43 (27). Anal. Calcd for $\mathrm{C}_{30} \mathrm{H}_{38} \mathrm{O}_{6}$ : C, $72.85 ; \mathrm{H}, 7.74$. Found: C, 73.02; H, 7.86 .

1,2-O-Isopropylidene-3-O-(p-phenylbenzyl)-4,5-O-[(1'S)-trans-2'-octen-1'-yl]- $\boldsymbol{\beta}$-d-fructopyranose (40B): $36 \%$ as an oily solid; $[\alpha]_{\mathrm{D}}-76.9^{\circ}\left(c 0.09\right.$, acetone); ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ) $\delta 0.87$ (t, 3 H ), 1.23-1.49 (m, 6 H$), 1.43$ ( $\mathrm{s}, 3 \mathrm{H}$ ), 1.49 ( s 3 H ), 3.50 (d, $J=7.54 \mathrm{~Hz}, 1 \mathrm{H}), 3.89(\mathrm{~d}, J=8.36 \mathrm{~Hz}, 1 \mathrm{H}), 4.09(\mathrm{~m}, 4 \mathrm{H})$, 4.54 (dd, $J=5.13 \mathrm{~Hz}, J=7.51 \mathrm{~Hz}, 1 \mathrm{H}), 4.73(\mathrm{~d}, J=11.96$ $\mathrm{Hz}, 1 \mathrm{H}$ ), 4.98 (d, $J=11.96 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.47 (m, 2 H ), 5.85 ( m , $1 \mathrm{H}), 7.32-7.60(\mathrm{~m}, 9 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) $13.98,22.46,26.08$, 26.89, 28.27, 31.39, 31.90, 60.27, 71.83, 72.56, 72.74, 73.65, $78.41,103.59,104.37,112.39,126.66,127.09,127.30,128.48$, 128.75, 136.93, 137.57 ; IR $2926 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) 494 ( $\mathrm{M}^{+}$), 312 ( 0.7 ), 282 ( 0.9 ), 254 (3), 211 (14), 167 (100), 125 (8), 69 (23). Anal. Caled for $\mathrm{C}_{30} \mathrm{H}_{38} \mathrm{O}_{6}$ : C, 72.85; H, 7.74. Found: C, 73.13; H, 7.68.

1,2-O-Isopropylidene-3- $O$ - $p$-phenylbenzyl)-4,5-O-[(1'R)-trans-3'-phenyl-2'-propen-1'-yl]- $\beta$-D-psicopyranose (41A): $38 \%$ as an oily solid; $[\alpha]_{D}-160.6^{\circ}$ (c 0.127, acetone); ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.46(\mathrm{~s}, 3 \mathrm{H}), 1.51(\mathrm{~s}, 3 \mathrm{H}), 3.75(\mathrm{~d}, J=3.04 \mathrm{~Hz}, 1$ H), 3.87 (m, 2 H), 4.04 (d, $J=9.22 \mathrm{~Hz}, 1 \mathrm{H}$ ), $4.20(\mathrm{dm}, J=$ $7.36 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.38 (d, $J=9.22 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.45 (dd, $J=3.18$ $\mathrm{Hz}, J=7.36 \mathrm{~Hz}, 1 \mathrm{H}), 4.83(\mathrm{~s}, 2 \mathrm{H}), 5.42(\mathrm{~d}, J=6.79 \mathrm{~Hz}, 1 \mathrm{H})$, 6.17 (dd, $J=6.62 \mathrm{~Hz}, J=15.97 \mathrm{~Hz}, 1 \mathrm{H}), 6.75(\mathrm{~d}, J=16.04$ $\mathrm{Hz}, 1 \mathrm{H}$ ), $7.23-7.60(\mathrm{~m}, 14 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 26.26,26.55$, $61.84,72.39,73.58,73.72,73.93,74.79,104.83,105.52,110.15$, $124.51,126.99,127.02,127.15,127.30,128.41,128.49,128.75$, $135.58,136.16,136.65,140.76$; IR 2989, $1662 \mathrm{~cm}^{-1}$; MS (EI) $m / e$ (relative intensity) $500\left(\mathrm{M}^{+}\right), 333(7), 256$ (4), 167 (100), 131 (23), 91 (5), 72 (11), 43 (12). Anal. Calcd for $\mathrm{C}_{31} \mathrm{H}_{32} \mathrm{O}_{6}$ : C, 74.38; H, 6.44. Found: C, 74.11; H, 6.32.

1,2-O-Isopropylidene-3-O-(p-phenylbenzyl)-4,5-O-[(1'S)-trans-3'-phenyl- $2^{\prime}$-propen- $1^{\prime}$-yl]- $\beta$-D-psicopyranose (41B): $36 \%$ as a white solid; $[\alpha]_{\mathrm{D}}-60.0^{\circ}$ (c 0.22 , acetone); mp 95.3 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.42(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 3 \mathrm{H}), 3.72(\mathrm{~d}, \mathrm{~J}=$ $3.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.96(\mathrm{~m}, 2 \mathrm{H}), 4.01(\mathrm{~d}, J=9.24 \mathrm{~Hz}, 1 \mathrm{H}), 4.24(\mathrm{~d}$, $J=9.23 \mathrm{~Hz}, 1 \mathrm{H}), 4.29(\mathrm{dm}, 1 \mathrm{H}), 4.54(\mathrm{dd}, J=3.29 \mathrm{~Hz}, J=$ $6.68 \mathrm{~Hz}, 1 \mathrm{H}), 4.83(\mathrm{~s}, 2 \mathrm{H}), 5.86(\mathrm{~d}, J=5.93 \mathrm{~Hz}, 1 \mathrm{H}), 6.10$ (dd, $J=5.91 \mathrm{~Hz}, J=15.85 \mathrm{~Hz}, 1 \mathrm{H}), 6.70(\mathrm{~d}, J=15.85 \mathrm{~Hz}, 1$ $\mathrm{H}), 7.24-7.60(\mathrm{~m}, 14 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) 26.36,26.51,62.32$, $72.40,73.31,73.43,73.67,75.36,105.22,105.47,110.60$, $125.95,126.92,127.09,127.21,127.35,128.31,128.56,128.77$, $134.08,135.76,136.56,140.75,140.95$; IR 2989, $1655 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $500\left(\mathrm{M}^{+}\right), 333$ (4), 256 (6), 217 (2), 167 (100), 131 (18), 91 (5), 72 (8), 43 (10). Anal. Calcd for $\mathrm{C}_{31} \mathrm{H}_{32} \mathrm{O}_{6}$ : C, $74.38 ; \mathrm{H}, 6.44$. Found: C, $74.15 ; \mathrm{H}, 6.50$.

1,2-O-Isopropylidene-3-O-benzyl-4,5-O-[3'-phenyl-2'-propyn-1'-yl]- $\boldsymbol{\beta}$-d-fructopyranose (44): $84 \%$ as colorless oil (endo:exo $=10: 1$ based on ${ }^{1} \mathrm{H}$ NMR); ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.41$ ( $\mathrm{s}, 3 \mathrm{H}$ ), 1.48 (s, 3 H ), $3.80(\mathrm{~d}, J=7.43 \mathrm{~Hz}, 1 \mathrm{H}$ ) 3.84 (d, $J=$ $8.59 \mathrm{~Hz}, 1 \mathrm{H}), 4.10(\mathrm{~d}, J=8.59 \mathrm{~Hz}, 1 \mathrm{H}), 4.13(\mathrm{~s}, 2 \mathrm{H}), 4.12$ (m, 1 H ), 4.49 (dd, $J=5.1 \mathrm{~Hz}, J=7.50 \mathrm{~Hz} 1 \mathrm{H}), 4.73$ (d, $J=$ $11.97 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.02 (d, $J=11.97 \mathrm{~Hz}, 1 \mathrm{H}), 5.91$ (s, 1 H$), 7.21-$ 7.51 (m, 10 H ).

1,2-O-Isopropylidene-3-O-benzyl-4,5-O-[(1'R)-cis-3'-phen-yl-2'-propen-1'-yl]- $\boldsymbol{\beta}$-d-fructopyranose (45). To a solution of the alkynal acetal 44 (endo:exo $=10: 1,350 \mathrm{mg}, 0.83 \mathrm{mmol}$ ) in methanol ( 8 mL ) was added Lindlar catalyst ( $35 \mathrm{mg}, 5 \%$ $\left.\mathrm{Pd} / \mathrm{CaCO}_{3} . \mathrm{PbO}\right)$. Then, a toy balloon filled with hydrogen was attached to the flask. The reaction mixture was stirred overnight at room temperature and then filtered through Celite. After evaporation of the solvent, the residue was purified by flash column chromatography on silica gel with 1:6:2 ether-hexane-chloroform to afford the cis-alkenal acetal 45 ( $287 \mathrm{mg}, 82 \%$ ) as a white solid: $[\alpha]_{\mathrm{D}}-14.5^{\circ}$ (c 0.069 , acetone); mp $79.7^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ) $\delta 1.42$ ( $\mathrm{s}, 3 \mathrm{H}$ ), 1.49 (s, $3 \mathrm{H}), 3.49(\mathrm{~d}, J=7.22 \mathrm{~Hz}, 1 \mathrm{H}), 3.87(\mathrm{~d}, J=8.41 \mathrm{~Hz}, 1 \mathrm{H})$, $4.07-4.14(\mathrm{~m}, 4 \mathrm{H}), 4.41(\mathrm{dd}, J=5.98 \mathrm{~Hz}, J=7.19 \mathrm{~Hz}, 1 \mathrm{H})$,
$4.68(\mathrm{~d}, J=12.04 \mathrm{~Hz}, 1 \mathrm{H}), 4.95(\mathrm{~d}, J=12.04 \mathrm{~Hz}, 1 \mathrm{H}), 5.63$ (d, $J=7.69 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.70 (dd, $J=7.65 \mathrm{~Hz}, J=10.54 \mathrm{~Hz}, 1$ $\mathrm{H}), 6.85(\mathrm{~d}, J=10.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.24-7.38(\mathrm{~m}, 10 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 26.04,26.92,59.88,71.85,72.93,76.04,76.40,77.37$, $99.92,104.38,112.4,127.63,127.92,128.22,128.27,128.98$, $135.5,136.15,138$; IR $3032,1649 \mathrm{~cm}^{-1}$; MS (EI) $m / e$ (relative intensity) $424\left(\mathrm{M}^{+}, 10\right), 333$ (11), 260 (10), 217 (4), 180 (7), 171 (21), 131 (22), 115 (23), 91 (100), 69 (7), 59 (12). Anal. Calcd for $\mathrm{C}_{25} \mathrm{H}_{28} \mathrm{O}_{6}$ : $\mathrm{C}, 70.74 ; \mathrm{H}, 6.65$. Found: $\mathrm{C}, 70.68 ; \mathrm{H}$, 6.71 .

1,2-O-Isopropylidene-3- $O$-(p-phenylbenzyl)-4,5-O-[6'-phenyl-2'-hexyn-1'-yl]- $\beta$-D-fructopyranose (46): $65 \%$ as a colorless oil; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.42(\mathrm{~s}, 3 \mathrm{H}), 1.48(\mathrm{~s}, 3 \mathrm{H})$, $1.86(\mathrm{~m}, 2 \mathrm{H}), 2.29(\mathrm{~m}, 2 \mathrm{H}), 2.71(\mathrm{t}, 2 \mathrm{H}), 3.77(\mathrm{~d}, J=7.46 \mathrm{~Hz}$, $1 \mathrm{H}), 3.88(\mathrm{~d}, J=8.42 \mathrm{~Hz}, 1 \mathrm{H}), 4.06(\mathrm{~m}, 4 \mathrm{H}), 4.44(\mathrm{dd}, J=$ $5.08 \mathrm{~Hz}, J=7.49 \mathrm{~Hz}, 1 \mathrm{H}), 4.73(\mathrm{~d}, J=12.03 \mathrm{~Hz}, 1 \mathrm{H}), 5.04$ $(\mathrm{d}, J=12.03 \mathrm{~Hz}, 1 \mathrm{H}), 5.69(\mathrm{t}, 1 \mathrm{H}), 7.12-7.58(\mathrm{~m}, 14 \mathrm{H}) ;{ }^{13} \mathrm{C}$ $\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) 18.18,26.06,29.91,29.70,34.79,59.56,71.73$, $73.27,75.33,76.04,77.86,87.78,93.25,104.49,112.34,126.01$, $127.04,127.07,127.26,128.22,128.41,128.44,128.75,137.29$, $140.53,140.88,141.15$; IR $2932 \mathrm{~cm}^{-1}$.

1,2-O-Isopropylidene-3- $O$-(p-phenylbenzyl)-4,5-O-[(1'R)-cis-6'-phenyl- $2^{\prime}$-hexen-1'-yl]- $\beta$-D-fructopyranose (47). To a solution of the acetylenic acetal 46 ( $800 \mathrm{mg}, 1.48 \mathrm{mmol}$ ) in methanol ( 20 mL ) was added Lindlar catalyst ( $80 \mathrm{mg}, 5 \% \mathrm{Pd} /$ $\mathrm{CaCO}_{3} \cdot \mathrm{PbO}$ ). Then, a toy balloon filled with hydrogen was attached to the flask. The reaction mixture was stirred overnight at room temperature and then filtered through Celite. After evaporation of the solvent, the residue was purified by flash column chromatography on silica gel with 1:2 ether-hexane to afford the endo acetal 47 ( $632 \mathrm{mg}, 79 \%$ ) as a white solid: $[\alpha]_{\mathrm{D}}-53.2^{\circ}$ (c 0.062 , acetone); $\operatorname{mp} 90.8^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 1.43(\mathrm{~s}, 3 \mathrm{H}), 1.49(\mathrm{~s}, 3 \mathrm{H}), 1.75(\mathrm{~m}, 2 \mathrm{H})$, $2.22(\mathrm{~m}, 2 \mathrm{H}), 2.63(\mathrm{t}, 2 \mathrm{H}), 3.45(\mathrm{~d}, J=7.09 \mathrm{~Hz}, 1 \mathrm{H}), 3.87(\mathrm{~d}$, $J=8.45 \mathrm{~Hz}, 1 \mathrm{H}), 4.08(\mathrm{~m}, 4 \mathrm{H}), 4.38(\mathrm{dd}, J=5.61 \mathrm{~Hz}, J=$ $7.14 \mathrm{~Hz} 1 \mathrm{H}), 4.68(\mathrm{~d}, J=12.06 \mathrm{~Hz}, 1 \mathrm{H}), 4.96(\mathrm{~d}, J=12.06$ $\mathrm{Hz}, 1 \mathrm{H}$ ), 5.48 (ddt, $J=8.02 \mathrm{~Hz}, J=10.69 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.66 (d, $J=8.02 \mathrm{~Hz}, 1 \mathrm{H}), 5.82(\mathrm{dt}, J=10.65 \mathrm{~Hz}, 1 \mathrm{H}), 7.14-7.59(\mathrm{~m}$, $14 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 26.08,26.91,27.32,31.04,35.27$, $59.90,71.88,72.67,75.92,76.56,77.10,99.37,104.38,112.27$, $125.79,127.03,127.08,127.25,128.34,128.44,128.74,137.17$, $137.51,141.98$; IR $2930 \mathrm{~cm}^{-1}$; MS (EI) $m / e$ (relative intensity) $542\left(\mathbf{M}^{+}, 2\right), 360(5), 302(2), 167$ (100), 131 (20), 117 (11), 91 (48), 81 (15). Anal. Calcd for $\mathrm{C}_{34} \mathrm{H}_{38} \mathrm{O}_{6}$ : $\mathrm{C}, 75.25 ; \mathrm{H}, 7.06$. Found: C, 75.47; H, 7.24.

General Procedure for Cyclopropanation of Acetals of $\alpha, \beta$-Unsaturated Aldehyde with Sugar Diol: Preparation of (1R,2R)-trans-1-(Hydroxymethyl)-2-phenylcyclopropane from 3-O-Benzyl-1,2-O-isopropylidene-4,5-O. [(1'R)-trans-3'-phenyl-2'-propen-1'-yl]- $\boldsymbol{\beta}$-d-fructopyranose (28A). To a solution of the trans-cinnamaldehyde acetal $28 A(150 \mathrm{mg}, 0.35 \mathrm{mmol})$ in 1,2-dichloroethane $(10 \mathrm{~mL})$ were added 1.0 M diethylzinc in hexane ( $1.77 \mathrm{~mL}, 1.77 \mathrm{mmol}$ ) and diiodomethane ( $0.29 \mathrm{~mL}, 3.5 \mathrm{mmol}$ ) at $-20^{\circ} \mathrm{C}$. The reaction mixture was stirred for 24 h at $0^{\circ} \mathrm{C}$, the reaction was quenched with saturated ammonium chloride solution, and then the solution was extracted with ethyl acetate. The organic layer was dried over magnesium sulfate, filtered, and evaporated. The resulting residue was purified by flash chromatography on silica gel with $1: 6: 2$ ether-hexanechloroform to give the cyclopropanated acetal ( $131 \mathrm{mg}, 84.8 \%$ ) as a colorless oil. A solution of the acetal ( $130 \mathrm{mg}, 0.3 \mathrm{mmol}$ ) in $80 \%$ acetic acid in $\mathrm{H}_{2} \mathrm{O}(3 \mathrm{~mL})$ was stirred for 15 min at 60 ${ }^{\circ} \mathrm{C}$ and then cooled to room temperature. The pH of the solution was adjusted to 7.5 with saturated $\mathrm{NaHCO}_{3}$, and then the solution was extracted with ethyl acetate. The organic layer was washed with water and saturated NaCl solution, dried over magnesium sulfate, and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel with $1: 7$ ethyl acetate-hexane to give $(1 R, 2 R)$ -trans-2-phenylcyclopropanecarboxaldehyde ( $40.3 \mathrm{mg}, 93 \%$ ) as a colorless oil. Further elution with 1:2 acetone-hexane gave the chiral auxiliary 3 - $O$-benzyl-1,2- $O$-isopropylidene- $\beta$-D-fructopyranose ( $74 \mathrm{mg}, 80.3 \%$ ) as a white solid. To a solution of the aldehyde ( $40 \mathrm{mg}, 0.27 \mathrm{mmol}$ ) in ethanol ( 3 mL ) was added sodium borohydride ( $15.5 \mathrm{mg}, 0.41 \mathrm{mmol}$ ) at room tempera-
ture. The reaction mixture was stirred for 1 h at room temperature, and the reaction was quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}$ solution. After evaporation of the solvent, the solution was diluted with water and ether. The organic layer was washed with water and saturated NaCl solution, dried over magnesium sulfate, and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel with $1: 4$ ethyl acetate-hexane to give ( $1 R, 2 R$ )-trans-1-(hydroxymethyl)-2-phenylcyclopropane ( $38.5 \mathrm{mg}, 95 \%$ ) as a colorless oil: $[\alpha]_{\mathrm{D}}-64.7^{\circ}$ (c $0.19, \mathrm{EtOH}$ as $71.3 \%$ ee) \{lit. ${ }^{11 c}$ $\left.[\alpha]_{\mathrm{D}}-92^{\circ}(c 1.23, \mathrm{EtOH})\right\} ;$ TLC $\left(20 \%\right.$ ethyl acetate/hexane) $R_{f}$ $0.11 ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 0.92(\mathrm{~m}, 2 \mathrm{H}), 1.45(\mathrm{~m}, 1 \mathrm{H}), 1.83(\mathrm{~m}$, $1 \mathrm{H}), 2.08(\mathrm{~b}, \mathrm{OH}), 3.60(\mathrm{dd}, J=6.71 \mathrm{~Hz}, J=2.24 \mathrm{~Hz}, 2 \mathrm{H})$, $7.16(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) ; 13.77,21.27,25.25,66.51$, $125.63,125.81,128.32,142.38$; IR $3350 \mathrm{~cm}^{-1}$; MS (EI) $m / e$ (relative intensity) $148\left(\mathrm{M}^{+}, 22\right), 130(21), 117$ (100), 104 (44), 91 (32), 77 (8), 51 (8).

The results with other sugar derivatives are summarized in Tables 4, 5, 6, and 9. The properties of pertinent compounds are listed below.
(1R,2R)-trans-2-((Benzyloxy)methyl)-1-(hydroxymethyl)cyclopropane: a colorless oil; $[\alpha]_{\mathrm{D}}-3.05^{\circ}\left(c 0.6, \mathrm{CHCl}_{3}\right.$ as $21 \%$ ee) $\left\{\mathrm{lit} .{ }^{11 \mathrm{~b}}[\alpha]_{\mathrm{D}}-6.0^{\circ}\right.$ (c $1.02, \mathrm{CHCl}_{3}$ as $35 \%$ ee $\left.)\right\} ;$ TLC ( $33 \%$ ethyl acetate/hexane) $R_{f} 0.15 ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) 0.43(\mathrm{t}, 2 \mathrm{H})$, $0.99(\mathrm{~m}, 2 \mathrm{H}), 1.64(\mathrm{~b}, 1 \mathrm{H}), 3.38(\mathrm{~m}, 4 \mathrm{H}), 4.51(\mathrm{~s}, 2 \mathrm{H}), 7.30$ (m, 5 H ); ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) 7.99, 16.75, 19.78, 66.20, 72.59, $73.46,127.59,127.68,128.36,141.10$; IR $3401 \mathrm{~cm}^{-1}$; MS (EI) $m / e$ (relative intensity) $192\left(\mathrm{M}^{+}\right), 161,129(2), 107$ (40), 91(100), 68(13).
(1R,2R)-trans-1-(Hydroxymethyl)-2-(2'-phenethyl)cyclopropane: a colorless oil; $[\alpha]_{\mathrm{D}}-17.1^{\circ}$ (c 1.5 as $52 \%$ ee) lit. ${ }^{17}[\alpha]_{\mathrm{D}}-24.6^{\circ}$ (c $1.3, \mathrm{CHCl}_{3}$ as $80 \%$ ee); TLC ( $25 \%$ ethyl acetate/hexane) $R_{f} 0.2 ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) 0.34(\mathrm{~m}, 2 \mathrm{H}), 0.59$ (m, 1 H ), $1.20(\mathrm{~b}, 1 \mathrm{H}), 1.52(\mathrm{~m}, 2 \mathrm{H}), 2.69(\mathrm{~m}, 2 \mathrm{H}), 3.38(\mathrm{~m}$, $2 \mathrm{H}), 7.20(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) 9.76,16.80,21.38,35.27$, $35.82,66.97,125.73,128.77,128.42,142.15$; IR $3402 \mathrm{~cm}^{-1}$; MS (EI) $m / e$ (relative intensity) $176\left(\mathrm{M}^{+}\right), 158(3), 129(34), 104$ (21), 91 (100), 65 (10).
(1R,2R)-trans-1-(Hydroxymethyl)-2-pentylcyclopropane: a colorless oil, $[\alpha]_{D}-24^{\circ}\left(c 0.1, \mathrm{CHCl}_{3}\right.$ as $64 \%$ ee); TLC ( $20 \%$ ethyl acetate/hexane) $R_{f} 0.25 ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 0.30$ (m, 2 H ), $0.58(\mathrm{~m}, 1 \mathrm{H}), 0.85(\mathrm{~m}, 4 \mathrm{H}), 1.24(\mathrm{~m}, 9 \mathrm{H}), 3.40(\mathrm{dd}$, $2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) 9.91,14.06,17.21,21.22,22.65,29.27$, $31.65,33.55,67.26$; IR $3354 \mathrm{~cm}^{-1}$.
(1R)-2,2-Dimethylcyclopropanecarboxylic Acid. To a solution of $1,2-O$-isopropylidene-3- $O$-( $p$-phenylbenzyl)-4,5- $O$ [( $\left.1^{\prime} R\right)$-3'-methyl-2'-buten-1'-yl]- $\beta$-D-fructopyranose (39A, 330 $\mathrm{mg}, 0.73 \mathrm{mmol}$ ) were added 1.0 M diethylzinc in hexane ( 3.65 $\mathrm{mL}, 3.65 \mathrm{mmol}$ ) and diiodomethane ( $0.61 \mathrm{~mL}, 7.3 \mathrm{mmol}$ ) at $-15{ }^{\circ} \mathrm{C}$. The reaction mixture was then stirred for 24 h at 0 ${ }^{\circ} \mathrm{C}$. The reaction was quenched with saturated aqueous $\mathrm{NH}_{4}$ Cl , and the mixture was extracted with ethyl acetate. The organic layer was dried over $\mathrm{MgSO}_{4}$, filtered, and evaporated. The resulting residue was purified by flash chromatography on silica gel with 1:6:2 ether-hexane-chloroform to give pure cyclopropanated acetal ( $274 \mathrm{mg}, 81 \%$ ) as a colorless oil, to which ( $200 \mathrm{mg}, 0.43 \mathrm{mmol}$ ) in ethyl acetate ( 20 mL ) at $-78^{\circ} \mathrm{C}$ ozone gas was passed. After the solution was stirred for 1 h at $23^{\circ} \mathrm{C}$, excess ozone was removed by flushing the system with nitrogen. The solvent was removed by evaporation, and the residue was purified to give pure 1,2-O-isopropylidene-3-$O$-( $p$-phenylbenzoyl)-5-O-((2',2'-dimethylcyclopropyl)carbonyl)-$\beta$-D-fructopyranose ( $169 \mathrm{mg}, 79 \%$ ) as a colorless oil after flash chromatography on silica gel with 1:3 ethyl acetate-hexane. To a solution of the ester fructopyranose ( $150 \mathrm{mg}, 0.3 \mathrm{mmol}$ ) in ethanol ( 2 mL ) was added aqueous $2 \mathrm{~N} \mathrm{NaOH}(0.6 \mathrm{~mL})$ at

[^4]rt , and the mixture was stirred for 1 h at rt . After evaporation of solvent, the pH of the solution was adjusted to 3 with dilute HCl and then extracted with ethyl acetate. The combined extracts were washed with water and brine, dried over $\mathrm{MgSO}_{4}$, and concentrated under reduced pressure. To the residue was added isopropyl ether. After filtration of resulting solid, the filtrate was evaporated to give 2,2 -dimethylcyclopropanecarboxylic acid ( $17.5 \mathrm{mg}, 51 \%$ ) as a colorless oil: [ $\alpha]_{\mathrm{D}}-85^{\circ}$ ( $c 0.1$, $\mathrm{CHCl}_{3}$ as $60 \%$ ee $)\left\{\right.$ lit. ${ }^{14} \mathrm{ent}$-form $\left.[\alpha]_{\mathrm{D}}+142^{\circ}\left(\mathrm{c} 1.01, \mathrm{CHCl}_{3}\right)\right\}$; ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 0.95$ (dd, $J=5.47 \mathrm{~Hz}, J=8.95 \mathrm{~Hz}, 1 \mathrm{H}$ ), $1.04(\mathrm{t}, 1 \mathrm{H}), 1.15(\mathrm{~s}, 3 \mathrm{H}), 1.23(\mathrm{~s}, 3 \mathrm{H}), 1.45(\mathrm{dd}, J=5.48 \mathrm{~Hz}$, $J=8.95 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) ; 18.72,22.86,24.35,26.74$, $27.00,179.62$; IR $3051,1692 \mathrm{~cm}^{-1}$.

Methyl (1R,2S)-cis- and (1S,2S)-trans-2-Phenylcyclopropanecarboxylates. According to the general procedure, a solution of 3-O-benzyl-1,2-O-isopropylidene-4,5-O-[(1R)-cis-$3^{\prime}$-phenyl-2'-propen-1'-yl]- $\beta$-D-fructopyanose ( $\mathbf{4 5}, 280 \mathrm{mg}, 0.66$ mmol ), 1.0 M diethylzinc in hexane ( $3.3 \mathrm{~mL}, 3.3 \mathrm{mmol}$ ), diiodomethane ( $0.55 \mathrm{~mL}, 6.6 \mathrm{mmol}$ ), and 1,2 -dichloroethane $(17 \mathrm{~mL})$ was stirred for 20 h at $0^{\circ} \mathrm{C}$. After the usual workup, the resulting residue was purified by flash chromatography on silica gel with 1:6:2 ether-hexane-chloroform to give a mixture of $2: 1$ of cyclopropanated cis-56 and trans- 57 compounds ( $276 \mathrm{mg} 96 \%$ ) as a colorless oil (based on ${ }^{1} \mathrm{H}$ NMR, cis $\mathrm{H}_{9}, \delta 2.35$; trans $\mathrm{H}_{9}, \delta 2.10$ ). To a mixture of 56 and 57 ( 250 $\mathrm{mg}, 0.57 \mathrm{mmol}$ ) was added $80 \% \mathrm{AcOH}$ in $\mathrm{H}_{2} \mathrm{O}(5 \mathrm{~mL})$. The reaction mixture was stirred for 1 h at $60^{\circ} \mathrm{C}$ and cooled to room temperature. The pH of the solution was adjusted to 7.5 with saturated $\mathrm{NaHCO}_{3}$ solution, and then the solution was extracted with ether. The organic layer was washed with water and brine, dried over $\mathrm{MgSO}_{4}$, and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel with 1:7 ethyl acetate-hexane to give a mixture of $2: 1$ of ( $1 R, 2 S$ )-cis- and ( $1 S, 2 S$ )-trans-2-phenylcyclopropanal ( $77.8 \mathrm{mg} 93 \%$ ) as a colorless oil (cis aldehyde $\delta$ 8.64 , trans aldehyde $\delta 9.30$ ). To a mixture solution of ( $1 R, 2 S$ )-cis- and ( $1 \mathrm{~S}, 2 \mathrm{~S}$ )-trans-2-phenylcyclopropanal ( $75 \mathrm{mg}, 0.51$ mmol ), sodium cyanide ( $252 \mathrm{mg}, 5.1 \mathrm{mmol}$ ), acetic acid ( 0.12 $\mathrm{mL}, 2.1 \mathrm{mmol})$, and methanol ( 10 mL ) was added manganese(IV) oxide ( $1.79 \mathrm{~g}, 20.5 \mathrm{mmol}$ ) at room temperature. The reaction mixture was stirred for 48 h at room temperature and then filtered through Celite. After evaporation of the solvent, the residue was purified by flash column chromatography on silica gel with 1:5 ether-hexane to give colorless oils of $(1 R, 2 S)$-cis product ( $50.1 \mathrm{mg}, 55 \%$ ) 58 and ( $1 S, 2 S$ )-trans product ( $25 \mathrm{mg}, 28 \%$ ) 59 , respectively.

Methyl (1R,2S)-cis-2-phenylcyclopropanecarboxylate: a colorless oil; $[\alpha]_{D}-29.7^{\circ}$ (c $0.175, \mathrm{CHCl}_{3}$ as $82 \%$ ee) \{lit. ${ }^{13}$ ent-form $[\alpha]_{D}+32.8^{\circ}$ (c 1.99, $\left.\mathrm{CHCl}_{3}\right)$; TLC ( $17 \%$ ethyl acetate/ hexane) $R_{f} 0.23$; ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ) $\delta 1.31(\mathrm{~m}, 1 \mathrm{H}), 1.69(\mathrm{~m}, 1$ $\mathrm{H}), 2.07(\mathrm{~m}, 1 \mathrm{H}), 2.56(\mathrm{q}, 1 \mathrm{H}), 3.41(\mathrm{~s}, 3 \mathrm{H}), 7.21(\mathrm{~m}, 5 \mathrm{H})$; MS (EI) m/e (relative intensity) 176 ( $\mathrm{M}^{+}, 14$ ), 144 (22), 115 (100), 91 (43), 77 (14), 63 (15).

Methyl (1S,2S)-trans-2-phenylcyclopropanecarboxylate: a colorless oil; $[\alpha]_{D}+173^{\circ}$ (c $0.185, \mathrm{CHCl}_{3}$ as $58 \%$ ee) $\left\{1 i \mathrm{l} .{ }^{13}\right.$
$\left.[\alpha]_{\mathrm{D}}+324.7^{\circ}\left(\mathrm{c} 1.24, \mathrm{CHCl}_{3}\right)\right\}$; TLC ( $17 \%$ ethyl acetate/hexane) $R_{f} 0.28 ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 1.30(\mathrm{~m}, 1 \mathrm{H}), 1.57(\mathrm{~m}, 1 \mathrm{H}), 1.89$ (m, 1 H ), $2.51(\mathrm{~m}, 1 \mathrm{H}), 3.70(\mathrm{~s}, 3 \mathrm{H}), 7.20(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) 16.99, 23.92, 26.26, 51.89, 126.21, 126.51, 128.47; IR $1728 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $176\left(\mathrm{M}^{+}, 43\right), 144$ (38), 117 (100), 91 (20), 59 (5).
( $1 R, 2 S$ )-cis-1-(Hydroxymethyl)-2-phenylcyclopropane. To a mixture of ( $1 R, 2 S$ )-methyl cis-2-phenylcyclopropanecarboxylate ( $17.6 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) in dried ether ( 1 mL ) was added 1.0 M diisobutylaluminum hydride in hexane $(0.4 \mathrm{~mL}$, 0.4 mmol ) at $0^{\circ} \mathrm{C}$. The reaction mixture was stirred for 2 h at $0^{\circ} \mathrm{C}$ and then poured onto water and ether. The organic layer was separated, dried over $\mathrm{MgSO}_{4}$, and concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel with 1:4 ethyl acetatehexane to afford ( $1 R, 2 S$ )-cis-1-(hydroxymethyl)-2-phenylcyclopropane ( $60,13.2 \mathrm{mg}, 89 \%$ ) as a colorless oil: $[\alpha]_{\mathrm{D}}-60.8^{\circ}(\mathrm{c}$ $0.09 \mathrm{CHCl}_{3}$ as $82 \%$ ee) $\left\{\right.$ lit. ${ }^{16}$ ent-form $[\alpha]_{\mathrm{D}}+39^{\circ}\left(\mathrm{c} 2.42 \mathrm{CHCl}_{3}\right.$ as $50 \%$ ee), lit. ${ }^{11 \mathrm{a}}[\alpha]_{\mathrm{D}}-52^{\circ}$ (c 1.3, EtOH as $70 \%$ ee) $\}$; TLC ( $20 \%$ ethyl acetate/hexane) $R_{f} 0.15 ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 0.87$ (q, 1 H ), 1.04 (m, 1 H ), 1.41 (b, 1 H ), 1.47 (m, 1 H$), 2.27$ ( $\mathrm{m}, 1$ H ), 3.24 (dd, $J=8.39 \mathrm{~Hz}, 1 \mathrm{H}), 3.46$ (dd, $J=11.66 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.21 (m, 5 H ); ${ }^{13} \mathrm{C} \mathrm{NMR} \mathrm{( } \mathrm{CDCl}_{3}$ ); 7.66, 20.69, 20.82, 62.90, 126.19, 128.27, 128.61, 138.20 ; IR $3370 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $148\left(\mathrm{M}^{+}, 31\right), 130(34), 117$ (100), 91 (61), 65 (15), 51 (20).
(1R,2S)-cis-1-(Hydroxymethyl)-2-(3'-phenylpropyl)cyclopropane: a colorless oil; $[\alpha]_{D}+18.2^{\circ}$ (c 0.77 , EtOH as $80.1 \%$ ee) $\left\{\right.$ lit. ${ }^{11 \mathrm{a}}[\alpha]_{\mathrm{D}}+19^{\circ}$ (c 0.7, EtOH as $81 \%$ ee) $\}$; TLC ( $20 \%$ ethyl acetate/hexane) $R_{f} 0.18 ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}\right) \delta-0.05(\mathrm{~m}, 1 \mathrm{H})$, $0.75(\mathrm{~m}, 1 \mathrm{H}), 0.89(\mathrm{~m}, 1 \mathrm{H}), 1.12(\mathrm{~m}, 1 \mathrm{H}), 1.14(\mathrm{~b}, \mathrm{OH}), 1.28$ $(\mathrm{m}, 1 \mathrm{H}), 1.50(\mathrm{~m}, 1 \mathrm{H}), 1.77(\mathrm{~m}, 1 \mathrm{H}), 2.63(\mathrm{~m}, 1 \mathrm{H}), 3.59(\mathrm{~m}$, 1 H ), $7.25(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ) $9.46,15.99,18.17,28.14$, $31.83,35.75,63.27,125.67,128.28,128.36$; IR $3402 \mathrm{~cm}^{-1}$; MS (EI) m/e (relative intensity) $190\left(\mathrm{M}^{+}, 1\right), 159$ (5), 131 (32), 104 (56), 91 (100), 77 (10).

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Supplementary Material Available: Global minimum structures of acetals $28 \mathrm{~A}, 28 \mathrm{~B}, 35 \mathrm{~A}, 35 \mathrm{~B}, 41 \mathrm{~A}, 41 \mathrm{~B}$, and 45 , obtained by force field calculation ( 8 pages). This material is contained in libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS; see any current masthead page for ordering information.

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    (1) (a) Kleschick, W. A.; Reed, M. W.; Bordner, J. J. Org. Chem. 1987, 52, 3168. (b) Abdallah, H.; Gree, R.; Carrie, R. Tetrahedron Lett. 1982, 23, 503. (c) Kunz, H.; Ruck, K. Angew. Chem., Int. Ed. Engl. 1993, 32, 336. (d) Quinkert, G.; Baier, H.; Friedhelm, A. Angew. Chem., Int. Ed. Engl. 1980, 19, 1029. (e) Ando, N.; Yamamoto, Y.; Oda, J.; Inouyo, Y. Synthesis 1978, 688. (f) Dietl, F.; Haunschild, J.; Merz, A. Tetrahedron 1985, 41, 1193. Jacobsen, E. N.; Mungall, W. S.; Schroder, G.; Sharpless, K. B. J. Am. Chem. Soc. 1988, 110, 1968. Mash, E. A.; Nelson, K. A. J. Am. Chem. Soc. 1985, 107, 8256. Mash, E. A.; Torok, D. S. J. Org. Chem. 1989, 54, 250 and references cited therein. (g) Katsuki, T.; Sharpless, K. B. J. Am. Chem. Soc. 1980, 102, 5976. Haruta, R.; Ishiguro, M.; Ikeda, N.; Yamamoto Y. J. Am. Chem. Soc. 1982, 104, 7667. Arai, I.; Mori, A.; Yamamoto, H. J. Am. Chem. Soc. 1985, 107, 8254. Mori, A.; Arai, I.; Yamamoto, H. Tetrahedron 1986, 42, 6447. (h) Denis, A. M.; Girard, C.; Conia, J. M. Synthesis 1972, 549. Johnson, C. R. Pure Appl. Chem. 1987, 59, 969. (i) Nishimura, J.; Kawabata N.; Furukawa, J. Tetrahedron 1969, 25, 2647. Roush, W. R.; Russo-Rodriguez, S. J. Org. Chem. 1987, 52, 603.
    (2) (a) Moser, W. R. J. Am. Chem. Soc. 1969, 91, 1135, 1141. (b) Nozaki, H.; Takaya, H.; Noyori, R. Tetrahedron 1968, 24, 3655. (c) Aratani, T.; Yoneyoshi, Y.; Negase, T. Tetrahedron Lett. 1975, 1707. (d) Wulfman, D. S.; Mc Gibboney, C. G.; Steffen, E. K.; Thinh, N. V.; McDaniel R. S. Tetrahedron 1976, 32, 1257 and reference citied therein. (e) Krieger, P. E.; Landgrebe, J. A. J. Org. Chem. 1978, 43, 4447. (f) Doyle, M. P.; Dorow, R. L.; Terpstra, J. W.; Rodenhous, R. A. J. Org. Chem. 1985, 50, 1663. (g) Monpert, A.; Martelli, J.; Gree, R.; Carrie, R. Tetrahedron Lett. 1981, 22, 1961; Nouv. J. Chim. 1983, 7, 345.
    (3) (a) Kawabata, N.; Nakagawa, T.; Nakao, T.; Yamashita, S. J. Org. Chem. 1977, 42, 3031 . (b) Mash, E. A.; Nelson, K. A. J. Am. Chem. Soc. 1985, 107, 8256. (c) Arai, I.; Mori, A.; Yamamoto, H. J. Am. Chem. Soc. 1985, 107, 8254.

[^1]:    (4) (a) Charette, A. B.; Cote, B.; Marcoux, J.-F. J. Am. Chem. Soc. 1991, 113, 8166. (b) Charette, A. B.; Cote, B. J. Org. Chem. 1993, 58, 933.
    (5) (a) Winstein, S.; Sonnenberg, J.; de Uries, L. J. Am. Chem. Soc. 1959, 81 , 6523. (b) Douben, W. G.; Berezin, G. Y. J. Am. Chem. Soc. 1963, 85, 468. (c) Denmark, S. E.; Edwards, J. P.; Wilson, S. R. J. Am. Chem. Soc. 1991, 113, 723. (d) Charette, A. B.; Jean-Francois, M. Tetrahedron Lett. 1993, 7157. (e) Mash, E. A.; Hemperly, S. B. J. Org. Chem. 1990, 55, 2055. (f) Mash, E. A.; Hemperly, S. B.; Nelson, K. A.; Heidt, P. C.; Deuson, S. A. J. Org. Chem. 1990, 55, 2045.
    (6) Takagi, Y.; Lim, G. J.; Tsuchiya, T.; Umezawa, S. J. Chem. Soc., Perkin Trans. 1 1992, 657.
    (7) Herscovici, J.; Antonakis, K. J. Chem. Soc., Chem. Commun. 1980, 561.

[^2]:    (8) Mori, A.; Arai, I.; Yamamoto, H. Tetrahedron 1986, 42, 6447.
    (9) Naef, D.; Seebach, D. Helv. Chim. Acta 1985, 68, 135.

[^3]:    (10) (a) Denmark, S. E.; Edwards, J. P. J. Org. Chem. 1991, 56, 6974. (b) Denmark, S. E.; Edwards, J. P.; Wilson, S. R. J. Am. Chem. Soc. 1991, 113, 723. (c) Miyano, S.; Hashimoto, H. Bull. Chem. Soc. Jpn. 1973, 46, 892. (d) Miyano, S.; Yamashita, J.; Hashimoto, H. Bull. Chem. Soc. Jpn. 1972, 45, 1946.
    (11) (a) Ukaji, Y.; Nishimura, M.; Fujisawa, T. Chem. Lett. 1992, 61. (b) Takahashi, H.; Yoshioka, M.; Ohno, M.; Kobayashi, S. Tetrahedron Lett. 1992, 33, 2575 . Inoue, Y.; Sugita, T.; Walborsky, H. M. Tetrahedron 1964, 20, 1695. Morikawa, T.; Sasaki, H.; Hanai, R.; Shibuya, A.; Taguchi, T. J. Org. Chem. 1994, 59, 97. (c) Evans, D. A.; Woerpel, K. A.; Hinman, M. M.; Faul, M. M. J. Am. Chem. Soc. 1991, 113, 726.
    (12) (a) Molander, G. A.; Etter, J. B. J. Org. Chem. 1987, 52, 3942. (b) Molander, G. A.; Harring, L. S. J. Org. Chem. 1989, 54, 3525.

[^4]:    (14) Graham, D. W.; Ashton, W. T.; Barash, L.; Brown, J. E.; Brown, R. D.; Canning, L. F.; Chen, A.; Springer, J. P.; Rogers, E. F. J. Med. Chem. 1987, 30, 1074.
    (15) Fritschi, H.; Leutenegger, U.; Pfaltz, A. Helv. Chim. Acta 1988, 71, 1553.
    (16) Aratani, T.; Nakanishi, Y.; Nozaki, H. Tetrahedron 1970, 26, 1675.
    (17) The $[\alpha]_{D}$ value of $(1 R, 2 R)$-trans -1 -(hydroxymethyl)- 2 -( $2^{\prime}$-phen ethyl)cyclopropane, which was cited in ref 32 b , was obtained from Dr Kobayashi at Sagami Chemical Research Center.

